

Cripple Creek and Victor Water Monitoring
Quality Assurance Project Plan and Field
Sampling Guidance for Grassy Valley Monthly
Monitoring

October 24, 2024~~February 27, 2023~~

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Appendices

- Appendix A - Example Chain of Custody Record
- Appendix B -Field Sampling Bottle Requirements and Holding Times
- Appendix C -Example Groundwater Field Sample Record Form
- Appendix D -Reagent Water Specifications
- Appendix E - Well Evacuation Calculation
- Appendix F - Low Flow Purge Volume Calculations~~YSI Pro Water Quality Meter Calibration Instructions~~
- Appendix G - YSI Pro Water Quality Meter Calibration Instructions~~CC&V Grassy Valley Sample Locations Identified Sample Names~~

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Appendix H – Field Level Risk Assessment~~CC&V Surface Water & Groundwater SOP~~

Appendix I – CC&V Surface Water Flume & Weir Calibration SOP

Appendix J – CC&V Sample Shipping SOP

Appendix K – Myron L Ultrameter II Calibration Procedure

Appendix L – Aqua Troll 600 Calibration Procedure

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1. Introduction

This Quality Assurance Project Plan (QAPP) for Grassy Valley monthly monitoring describes the quality assurance and quality control (QA/QC) procedures and practices for this location at Newmont’s Cripple Creek and Victor Mine (CC&V). Included are standard operating procedures (SOP) for the various stages of sample collection, shipping and analyses. Sample schedules, location maps and analytical requirements are also included. This is a “living” document and is intended as guidance for site personnel; the QAPP is updated as needed to reflect changes to CC&V’s water monitoring program and regulatory requirements. If a change to the QAPP is required, it will be implemented immediately and revised accordingly based on regulatory consultation as needed.

Accurate water quality data are critical to ensure permit compliance and demonstrate that water resources are not impacted by operations. The generation of reliable data begins with the collection of the sample. Adherence to the SOPs will ensure that samples are representative, and collected in accordance with standard water sampling methods and QA/QC protocols. To produce data of defensible quality, this quality control program will be strictly adhered to during sample collection.

The water-sampling program includes collecting samples, recording field data, submitting samples for analyses, reviewing and recording analytical results.

1.1. Program Organization

Duties of key program personnel are listed in Table 1 below:

Table 1. Summary of Key Personnel

Name	Role	Contact Info
------	------	--------------

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Jordan Cranford	Environmental Technician	Jordan.Cranford@newmont.com
SVL	Contract Laboratory	TBD
Heather Lapierre	Contract Laboratory Director	heather@svl.net (208) 783-1286

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- **Senior Environmental Manager:** The Senior Environmental Manager ensures the overall QA/QC program development and implementation. The Senior Environmental Manager allocates resources to ensure QA/QC and compliance criteria are met.
- **Site Water Coordinator:** The Site Water Coordinator is the program technical expert. The Site Water Coordinator oversees resource allocation, program implementation, coordinates field efforts, ensures sampling schedules are met and manages laboratory sub-contracts.
- ~~Senior Environmental Technicians~~ **Technician:** The ~~Senior Environmental Technicians~~ **Technician** executes the QAPP as the sampler in the field, ensuring program adherence during sample collection and shipment.
 - **Contract Laboratory Manager:** Ensure analyses of environmental samples are conducted in adherence with regulatory, industry and program QA/QC requirements.

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1.2. Program Objectives

The general objectives of the environmental monitoring and sampling program are:

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- Collect, whenever possible, representative samples and document instances when sampling conditions are not ideal
- Ensure environmental samples are collected and analyzed according to regulatory and program QA/QC requirements
- Provide a record and insight of natural variability in environmental data as a

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function of seasonal meteoric changes and site evolution as influenced by historic mining practices

- Document and record environmental data history
- Identify potential environmental impacts from site activities

2.0 SAMPLING PLAN

1.3.

~~Monitoring within Grassy Valley consists of the following: The following locations are recognized as Grassy Valley Monthly surface monitoring locations: one location on the southern toe of the ECOSA facility (SEEP 1), one location on the toe of the ECOSA Facility mid-way between the northern and southern extents of the facility (SEEP 2) and one location on Grassy Creek in Lower Grassy Valley (GV-06). Samples collected at the surface water monitoring stations, shown in Table 2.1. A list of surface water parameters is provided below in Table 2.2. This program monitors water quality at the toe of the ECOSA facility and downgradient surface water quality below the ECOSA facility.~~

- ~~• Monthly sampling/monitoring of 5 surface water locations~~
- ~~• Monthly sampling/monitoring of 34 groundwater locations~~
- ~~• Monthly sampling/monitoring of 5 stormwater detention ponds (i.e. EMP's)~~
- ~~• Monthly sampling/monitoring of seepage locations~~

~~Reporting frequencies and monitoring frequencies, locations, and parameters may be modified with the approval of the Colorado Division of Reclamation Mining and Safety (DRMS). If the monitoring or reporting requirements are modified, CC&V will update this QAPP accordingly.~~

2.0 Surface Water

2.1 Surface water sampling locations

CC&V collects surface water samples from ~~five~~³ locations within Grassy Valley. Table 2.1 below presents the Grassy Valley Monthly monitoring compliance surface water monitoring locations. ~~The GV-06 location is considered the point-of-compliance. ,and~~

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Table 2.2 below depicts surface water compliance monitoring location water monitoring parameters. Table 2.3 below presents compliance limitations as specified in Colorado Regulation 32, and Figure 2.1 presents a map of the Grassy Valley monthly surface water monitoring locations.

Table 2.1 ***Surface Water Monitoring Locations*** ~~Grassy Valley Monthly surface water monitoring locations~~

Site Number	Location	Monitoring Frequency
GV-06	Grassy Creek - Lower Grassy Valley	Monthly
GV-4.5	Grassy Creek - Lower Grassy Valley beyond affected lands boundary	Monthly
GV-0506	Grassy Creek - Lower Grassy Valley upstream of the confluence with Beaver Creek	Monthly
GV-02	Grassy Creek - Upper Grassy Valley	Monthly
GV-03	Grassy Creek - Middle Grassy Valley	Monthly

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Table 2.2 ~~Monthly Grassy Valley~~ ***Surface Water Monitoring Parameters***

Parameters		
pH (Field)	Barium (mg/L) Total Recoverable	Temperature (°C)
Ammonia (mg/L as N) Total	Beryllium (mg/L) Total Recoverable	Manganese (mg/L) Total Recoverable
Cyanide [FREE] (Dissolved)	Cadmium (mg/L) Dissolved	Manganese (mg/L) Dissolved

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▲ Fluoride (mg/L)	Cadmium (mg/L) Total Recoverable	Mercury (mg/L) Total Recoverable
▲ Nitrate (mg/L as N)	Chlorine (mg/L)	Molybdenum (mg/L) Total Recoverable
▲ Nitrite (mg/L as N)	Chromium (mg/L) Dissolved	Nickel (mg/L) Dissolved
▲ Boron (mg/L)	Chromium III (mg/L) Dissolved	Nickel (mg/L) Total
▲ Chloride (mg/L)	Chromium III (mg/L) Total	Phosphorus (mg/L)
▲ Sulfate (mg/L)	Chromium VI (mg/L) Dissolved	Selenium (mg/L) Dissolved
▲ Aluminum (mg/L) Dissolved	Copper (mg/L) Dissolved	Silver (mg/L) Dissolved
▲ Cyanide [WAD]	Iron (mg/L) Total Recoverable	Sulfide (mg/L)
▲ Dissolved Oxygen (mg/L)	Iron (mg/L) Dissolved	Thallium (mg/L) Dissolved
▲ Antimony (mg/L) Total Recoverable	Lead (mg/L) Total Recoverable	Uranium (mg/L) Dissolved
▲ Arsenic (mg/L) Total Recoverable	Lead (mg/L) Dissolved	Zinc (mg/L) Dissolved
▲ Arsenic (mg/L) Dissolved		

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Table 2.3 Grassy Valley Surface Water Standards~~water standards~~

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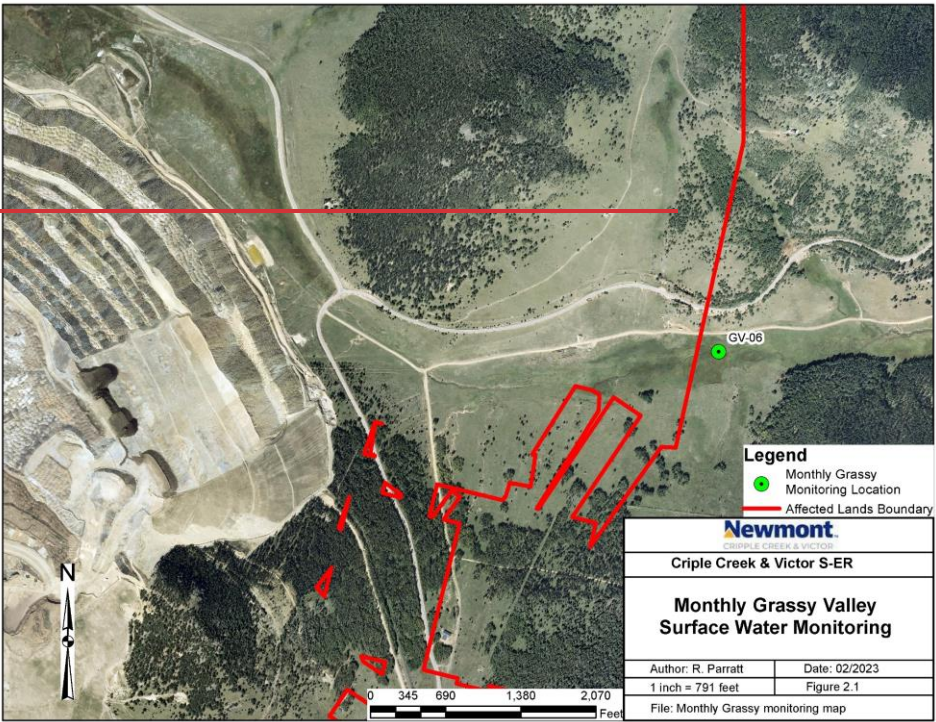
24. Mainstem of East and West Beaver Creeks, including all tributaries and wetlands, from the source to the confluence with Beaver Creek; mainstem of Beaver Creek from the source to the point of diversion to Brush Hollow Reservoir.						
COARUA24	Classifications	Physical and Biological			Metals (ug/L)	
Designation	Agriculture	DM	MWAT		acute	chronic
Reviewable	Aq Life Cold 1	Temperature °C	CS-II CS-II	Arsenic	340	---
	Recreation E		acute chronic	Arsenic(T)	---	0.02
	Water Supply	D.O. (mg/L)	6.0	Cadmium	TVS	TVS
Qualifiers:		D.O. (spawning)	---	Cadmium(T)	5.0	---
Other:		pH	6.5 - 9.0	Chromium III	---	TVS
Temporary Modification(s):		chlorophyll a (mg/m ²)	---	Chromium III(T)	50	---
Arsenic(chronic) = hybrid		E. coli (per 100 mL)	---	Chromium VI	TVS	TVS
Expiration Date of 12/31/2024			126	Copper	TVS	TVS
*Uranium(acute) = See 32.5(3) for details.				Iron	---	WS
*Uranium(chronic) = See 32.5(3) for details.				Iron(T)	---	1000
		Inorganic (mg/L)		Lead	TVS	TVS
		acute chronic		Lead(T)	50	---
		Ammonia	TVS TVS	Manganese	TVS	TVS/WS
		Boron	---	Mercury(T)	---	0.01
		Chloride	---	Molybdenum(T)	---	150
		Chlorine	0.019 0.011	Nickel	TVS	TVS
		Cyanide	0.005	Nickel(T)	---	100
		Nitrate	10	Selenium	TVS	TVS
		Nitrite	---	Silver	TVS	TVS(tr)
		Phosphorus	---	Uranium	varies*	varies*
		Sulfate	---	Zinc	TVS	TVS
		Sulfide	---			

24. Mainstem of East and West Beaver Creeks, including all tributaries and wetlands, from the source to the confluence with Beaver Creek; mainstem of Beaver Creek from the source to the point of diversion to Brush Hollow Reservoir.						
COARUA24	Classifications	Physical and Biological			Metals (ug/L)	
Designation	Agriculture	DM	MWAT		acute	chronic
Reviewable	Aq Life Cold 1	Temperature °C	CS-II CS-II	Arsenic	340	---
	Recreation E		acute chronic	Arsenic(T)	---	0.02
	Water Supply	D.O. (mg/L)	---	Cadmium	TVS	TVS
Qualifiers:		D.O. (spawning)	---	Cadmium(T)	5.0	---
Other:		pH	6.5 - 9.0	Chromium III	---	TVS
Temporary Modification(s):		chlorophyll a (mg/m ²)	---	Chromium III(T)	50	---
Arsenic(chronic) = hybrid		E. coli (per 100 mL)	---	Chromium VI	TVS	TVS
Expiration Date of 12/31/2024			126	Copper	TVS	TVS
*Uranium(acute) = See 32.5(3) for details.				Iron	---	WS
*Uranium(chronic) = See 32.5(3) for details.				Iron(T)	---	1000
		Inorganic (mg/L)		Lead	TVS	TVS
		acute chronic		Lead(T)	50	---
		Ammonia	TVS TVS	Manganese	TVS	TVS/WS
		Boron	---	Mercury(T)	---	0.01
		Chloride	---	Molybdenum(T)	---	150
		Chlorine	0.019 0.011	Nickel	TVS	TVS
		Cyanide	0.005	Nickel(T)	---	100
		Nitrate	10	Selenium	TVS	TVS
		Nitrite	---	Silver	TVS	TVS(tr)
		Phosphorus	---	Uranium	varies*	varies*
		Sulfate	---	Zinc	TVS	TVS
		Sulfide	---			

Figure 2.1 Surface Water Monitoring Locationswater monitoring locations

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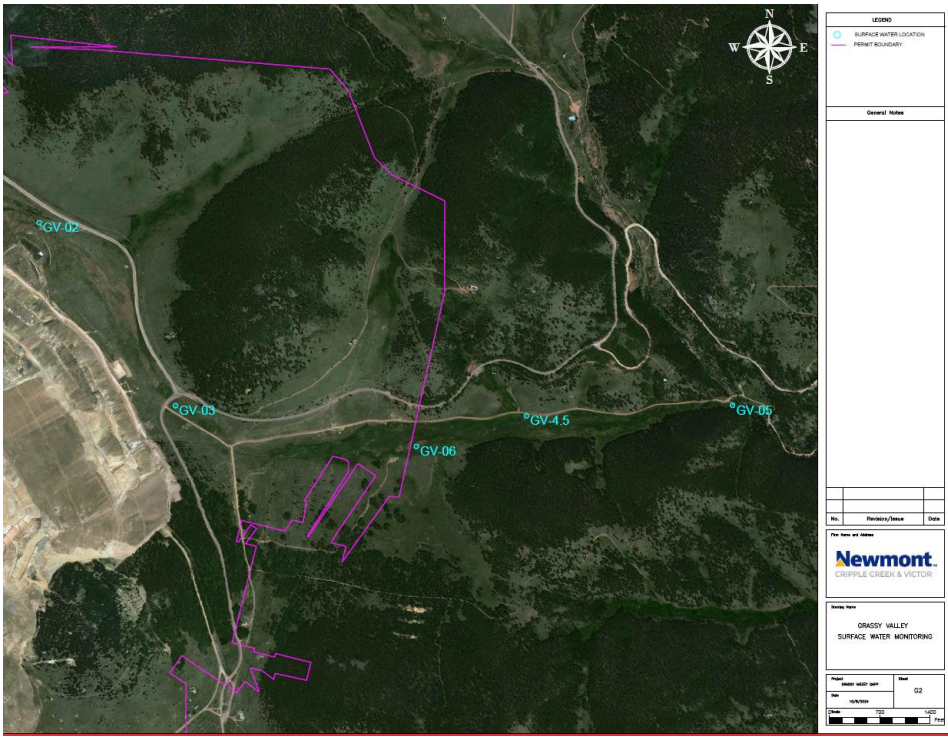


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3.0 Monthly Grassy Valley Monitoring Groundwater
Monitoring Locations

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Monthly Grassy Valley groundwater monitoring locations within Grassy Valley evaluatefor water quality downgradient of CC&V's ECOSA facility. The GVMW-26A and GVMW-26B are considered the points-of-compliance. Table 3.1 below presents the groundwater monitoring locationslocation for the monthly Grassy Valley monitoringgrassy valley sampling, Table 3.2 presents the constituents which this location is sampled for, Table 3.3 presents the standards to which this groundwater monitoring location is evaluated against, and Figure 3.1 shows where the groundwater monitoring location is in Grassy Valley.

Table 3.1. Groundwater Monitoring Locationsgroundwater monitoring locations

Site Number	Longitude (Dec. Deg) <u>Location</u>	Latitude (Dec. Deg)	Monitoring Frequency
GVMW-4A	-105.1340	38.7505	Monthly
GVMW-7A	-105.1244	38.7477	Monthly
GVMW-7B	-105.1244	38.7477	Monthly
GVMW-8A	-105.1202	38.7413	Monthly
GVMW-8B	-105.1202	38.7413	Monthly
GVMW-10	-105.1252	38.7444	Monthly
GVMW-15A	-105.1358	38.7495	Monthly
GVMW-15B	-105.1357	38.7494	Monthly
GVMW-15C	-105.1359	38.7495	Monthly
GVMW-22A	-105.1110	38.7408	Monthly
GVMW-22B	-105.1111	38.7408	Monthly
GVMW-24A	-105.1187	38.7378	Monthly
GVMW-24B	-105.1187	38.7378	Monthly
GVMW-25	-105.1196Grassy Valley~ Downgradient of ECOSA	38.7403	Monthly
GVMW-26A26	-105.1110Grassy Valley	38.7396	Monthly

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<u>GVMW-26BSeep 1</u>	<u>-105.1111South toe of ECOSA</u>	<u>38.7396</u>	Monthly	
<u>GVMW-27Seep 2</u>	<u>-105.1217Toe of ECOSA</u>	<u>38.7412</u>	Monthly	
<u>GVMW-28</u>	<u>-105.1230</u>	<u>38.7422</u>	<u>Monthly</u>	
<u>GVMW-29</u>	<u>-105.1214</u>	<u>38.7389</u>	<u>Monthly</u>	
<u>GVMW-30</u>	<u>-105.1217</u>	<u>38.7395</u>	<u>Monthly</u>	
<u>GVMW-31</u>	<u>-105.1221</u>	<u>38.7401</u>	<u>Monthly</u>	
<u>GVMW-32</u>	<u>-105.1225</u>	<u>38.7406</u>	<u>Monthly</u>	
<u>GVMW-33</u>	<u>-105.1230</u>	<u>38.7411</u>	<u>Monthly</u>	
<u>GVMW-34</u>	<u>-105.1292</u>	<u>38.7471</u>	<u>Monthly</u>	
<u>GVMW-35A</u>	<u>-105.1301</u>	<u>38.7473</u>	<u>Monthly</u>	
<u>GVMW-35B</u>	<u>-105.1305</u>	<u>38.7475</u>	<u>Monthly</u>	
<u>GVMW-36</u>	<u>-105.1337</u>	<u>38.7486</u>	<u>Monthly</u>	
<u>GVMW-37A</u>	<u>-105.1181</u>	<u>38.7401</u>	<u>Monthly</u>	
<u>GVMW-37B</u>	<u>-105.1181</u>	<u>38.7402</u>	<u>Monthly</u>	
<u>OSABH-12</u>	<u>-105.1187</u>	<u>38.7377</u>	<u>Monthly</u>	
<u>OSABH-14</u>	<u>-105.1203</u>	<u>38.7414</u>	<u>Monthly</u>	
<u>OSABH-16</u>	<u>-105.1254</u>	<u>38.7445</u>	<u>Monthly</u>	
<u>OSABH-17</u>	<u>-105.1328</u>	<u>38.7483</u>	<u>Monthly</u>	
<u>OSABH-18</u>	<u>-105.1359</u>	<u>38.7495</u>	<u>Monthly</u>	

Table 3.2 Monthly Grassy Valley Groundwater Monitoring Parameters

Parameters		
Aluminum (dissolved)	Cyanide [FREE]	Nitrite (NO2)
Antimony (dissolved)	Fluoride (dissolved)	pH
Arsenic (dissolved)	Iron (dissolved)	Selenium (dissolved)
Barium (dissolved)	Lead (dissolved)	Silver (dissolved)
Beryllium (dissolved)	Lithium (dissolved)	Sulfate (dissolved)
Boron (dissolved)	Manganese (dissolved)	Thallium (dissolved)
Cadmium (dissolved)	Mercury (inorganic) (dissolved)	Total Nitrate + Nitrite (NO ₂ +NO ₃ -N)
Chloride (dissolved)	Molybdenum (dissolved)	Uranium (dissolved)
Chromium (dissolved)	Nickel (dissolved)	Vanadium (dissolved)
Cobalt (dissolved)	Nitrate (NO3)	Zinc (dissolved)
Copper (dissolved)	Cyanide [WAD] ¹	-

1 Weak Acid Dissociable

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Table 3.3 Grassy Valley Groundwater Standards

Parameter	mg/L	Parameter	mg/L
Aluminum	7	Manganese	3
Antimony	0.006	Mercury	0.002
Arsenic	0.01	Molybdenum	0.21
Barium	2	Nickel	0.2
Beryllium	0.004	Nitrate-as-N	10
Boron	0.75	Nitrite + Nitrate-as-N	11
Cadmium	0.005	Nitrite-as-N	1
Chloride	250	pH	6.0–8.5
Chromium	0.1	Selenium	0.024
Cobalt	0.05	Silver	0.05
Copper	0.2	Sulfate (GVMW-8A)	250
Cyanide [FREE]	0.2	Sulfate	250
Cyanide [WAD]	0.2	Thallium	0.002
Fluoride	2	Uranium	0.03
Iron	14	Vanadium	0.1
Lead	0.05	Zinc	2
Lithium	2.5		

Cells Highlighted in blue are Numeric Protection Limits

Cells Highlighted in white are Table Value Standards

Description	GVMW- 8A	Sitewide NPL (Existing)	Reg. 41 Table Value Standard
Aluminum (dis) (mg/L)	-	7	5
Antimony (dis) (mg/L)	-	-	0.006
Arsenic (dis) (mg/L)	-	-	0.01

<u>Barium (dis) (mg/L)</u>	-	-	<u>2</u>
<u>Beryllium (dis) (mg/L)</u>	-	-	<u>0.004</u>
<u>Boron (dis) (mg/L)</u>	-	-	<u>0.75</u>
<u>Cadmium (dis) (mg/L)</u>	-	<u>0.005</u>	<u>0.005</u>
<u>Chloride (dis) (mg/L)</u>	-	-	<u>250</u>
<u>Chromium (dis) (mg/L)</u>	-	-	<u>0.1</u>
<u>Cobalt (dis) (mg/L)</u>	-	-	<u>0.05</u>
<u>Copper (dis) (mg/L)</u>	-	<u>0.2</u>	<u>0.2</u>
<u>Cyanide [Free] (mg/L)</u>	-	-	<u>0.2</u>
<u>Cyanide [WAD] (mg/L)</u>	-	<u>0.2</u>	<u>---</u>
<u>Fluoride (mg/L)</u>	-	<u>2</u>	<u>2</u>
<u>Iron (dis) (mg/L)</u>	-	<u>14</u>	<u>0.3</u>
<u>Lead (dis) (mg/L)</u>	-	-	<u>0.05</u>
<u>Lithium (dis) (mg/L)</u>	-	-	<u>2.5</u>
<u>Manganese (dis) (mg/L)</u>	<u>1</u>	<u>3</u>	<u>0.05</u>
<u>Mercury (dis) (mg/L)</u>	-	<u>0.002</u>	<u>0.002</u>
<u>Molybdenum (dis) (mg/L)</u>	-	-	<u>0.21</u>
<u>Nickel (dis) (mg/L)</u>	-	<u>0.2</u>	<u>0.1</u>
<u>Nitrate as Nitrogen (mg/L)</u>	-	<u>10</u>	<u>10</u>
<u>Nitrite as Nitrogen (mg/L)</u>	-	<u>1</u>	<u>1</u>
<u>Nitrate + Nitrite as Nitrogen (mg/L)</u>	-	<u>11</u>	<u>10</u>
<u>pH Field - Upper</u>	<u>8.5</u>	<u>8.5</u>	<u>8.5</u>
<u>pH Field - Lower</u>	<u>6.5</u>	<u>6</u>	<u>6.5</u>
<u>Selenium (dis) (mg/L)</u>	-	<u>0.024</u>	<u>0.02</u>
<u>Silver (dis) (mg/L)</u>	-	-	<u>0.05</u>
<u>Sulfate (mg/L)</u>	<u>250</u>	-	<u>250</u>
<u>Thallium (dis) (mg/L)</u>	-	-	<u>0.002</u>
<u>Uranium (dis) (mg/L)</u>	-	-	<u>0.03</u>
<u>Vanadium (dis) (mg/L)</u>	-	-	<u>0.1</u>
<u>Zinc (dis) (mg/L)</u>	-	<u>2</u>	<u>2</u>

Note:

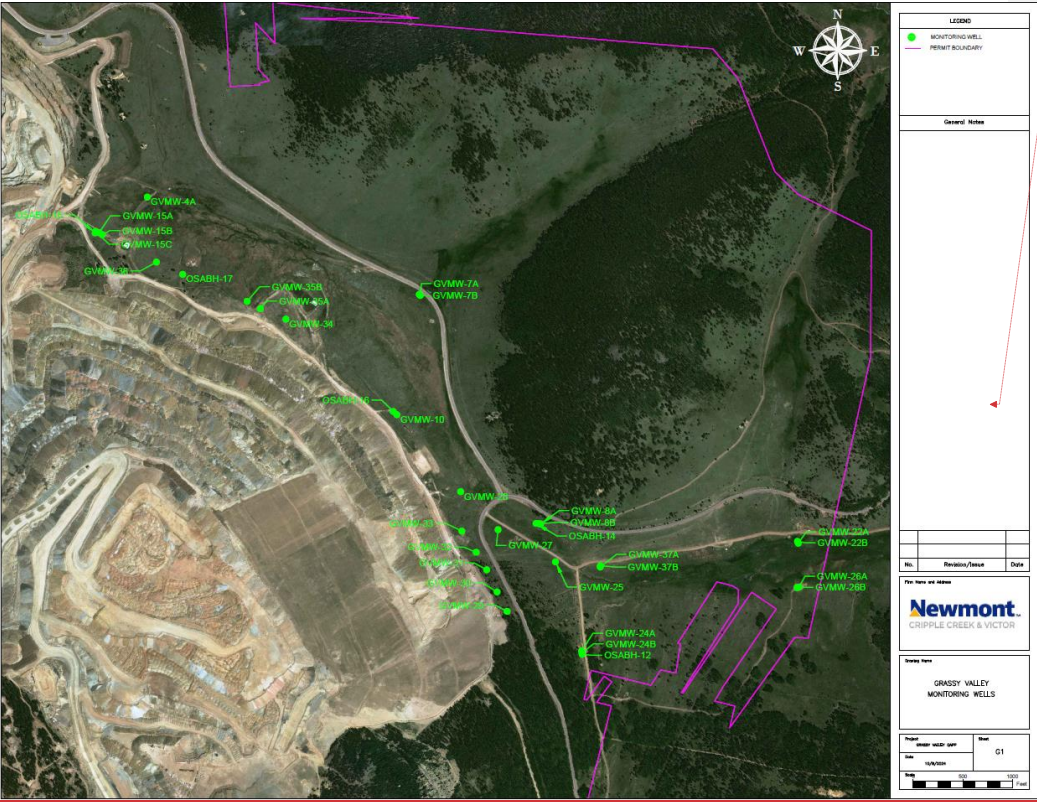
- If no NPL exists (Sitewide or well-specific) for that parameter the most stringent value from Colorado Regulation 41 Tables 1, 2, or 3 is applied
- dis – dissolved
- mg/L – milligram per liter

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Figure 3.1 Groundwater Monitoring Locations

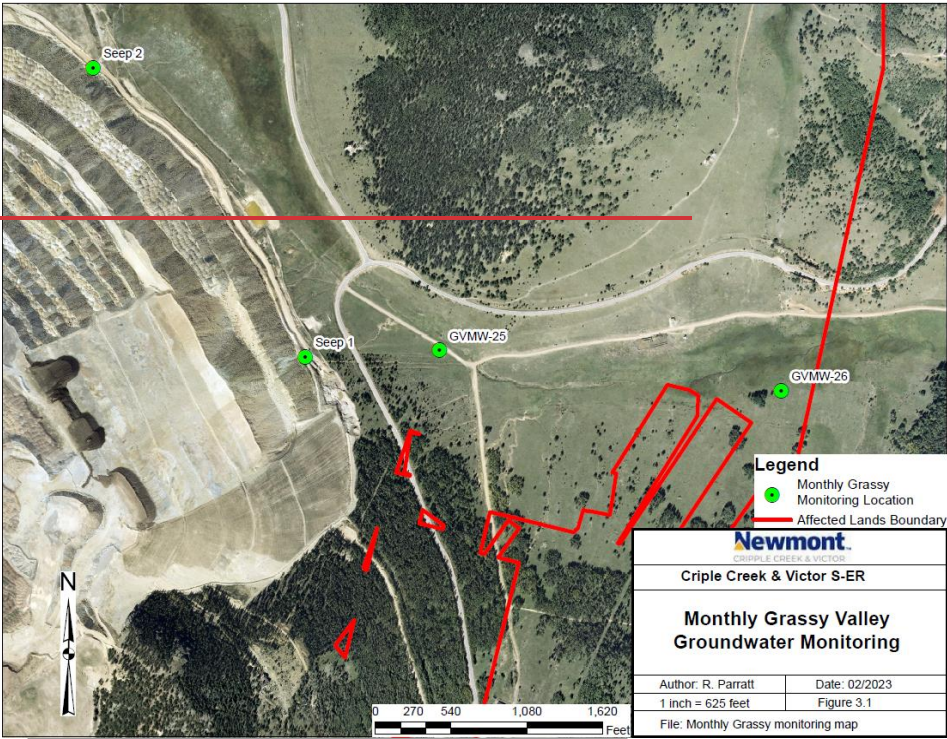
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*Location for GVMW-26 is approximate until exact coordinates are known

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4.0 EMPs

Enhanced Management Practices (EMPs) are stormwater collection/detention ponds that periodically have standing water. Table 4.1 below presents the EMP monitoring locations for the monthly monitoring. EMPs are sampled for the same constituents as the groundwater locations and compared to the same standards as the groundwater locations. Figure 4.1 shows the locations of the EMPs within Grassy Valley. EMPs will be sampled monthly given that they can be safely accessed, there is an adequate amount of water to be sampled, and the water is not frozen.

Table 4.1. EMP Locations

Site Number	Monitoring Frequency
EMP-16	Monthly
EMP-17	Monthly
EMP-17A	Monthly
EMP-17B	Monthly
EMP-20	Monthly

Figure 4.1 EMP Monitoring Locations



5.0 Seeps

Figure 5.1 shows the locations of the Seeps within Grassy Valley. Field parameters (pH, temperature, and conductivity) will be collected from the seep locations monthly. Samples will be collected on a quarterly basis from the seep locations and analyzed for the same parameters as the groundwater locations and compared to the same standards as the groundwater locations. Quarterly sampling of the seep locations will only occur if they can be safely accessed, there is an adequate amount of water to be sampled, and the water is not frozen.

Figure 5.1 Seep Locations



4.6.0 Field Technician Duties

Collection of reliable data and maintenance of analytical data are the foundation of compliance activities. Thus, the duties performed by the sampler provide the most critical element of the Environmental Department's efforts. Although this document primarily is focused on sample collection and handling methods, the generation of water data can be envisioned as a loop that includes more than simply sample collection. For any given sample, the sampler's duties have not been completed until this loop is closed. In general, a complete loop includes the collection of the sample, transmittal of the sample to a lab, receipt, review and, storage of analytical data.

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4.1.6.1 Tasks

Each of these steps includes several tasks, each of which must be conducted in accordance with the procedures outlined in this document. Specific duties include the following:

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- Collection of samples;
- Collection of duplicate and control samples;
- Collection of field data;
- Maintenance of equipment;
- Calibration of equipment used to collect field data;
- Tracking sample status;
- Data review and management;
- ~~Review of invoices and coding for payment;~~
- Updating sample schedules, maps and other documents as needed;
- Conducting periodic inventory of equipment;

Site Inspection

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6.2

The primary function of a sampler is to collect samples. Samplers should strive to be observant of environmental conditions while in the field and should be aware of circumstances or occurrences, which are unusual or different from past events. Leaks or damp areas, materials stored in possibly unauthorized places, wildlife in the vicinity of ponds or tailing impoundments are examples of things to be noted. Any concerns noted should be promptly brought to the attention of the area environmental coordinator.

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4.2.6.3 Quality Control

Newmont's quality control program consists of the following elements: sampler competence, utilization of standards, field blanks and duplicates, calibration of meters, equipment maintenance and routine auditing of sampling procedures. Analytical results of control samples will not be used to modify any sample analyses reports.

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4.3.6.4 Training

To ensure samples are collected and managed accordingly, the approved SOPs and regulatory requirements, samplers are trained by department personnel who are knowledgeable and experienced in Newmont's monitoring program. Samplers will be familiar with site history and conditions and will maintain active Mine Safety and Health Administration (MSHA) certifications. The contracted laboratory will be required to maintain appropriate certifications as needed.

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4.4.6.5 Calibration

All calibration ~~and calibration check~~ data will be documented ~~in the field log book.~~ All field equipment will be calibrated prior to field use on daily calibration forms. Calibration procedures shall follow the manufacturers' specifications. ~~A calibration check will be performed after all samples have been collected for the day. Calibration checks will not be used to correct pH readings taken during the day.~~

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5. Monthly Grassy Sampling Report Frequency

5.1. Sample Frequency

~~Per Technical Revision 132, surface water and groundwater samples are collected and reported to the Colorado Division of Reclamation and Mining Safety (DRMS). CC&V collects and submits surface and groundwater samples, duplicates and rinse blanks to our contracted laboratory for analysis.~~

Surface water, groundwater and QA/QC samples are collected monthly.

7.0

7.1 ~~5.2~~ Monthly Sampling Report

Monthly sampling reports are to be submitted to DRMS no later than 28 days after the last day of the preceding month. For example, reports are due on March 28 for February sampling event.

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6.8.0 CC&V Monthly Grassy Sample Types

6.1.8.1 Duplicate Samples

Duplicate samples are two or more samples collected at the same time from the same location and are used to check the analyzing laboratory's accuracy. For CC&V's monthly Grassy sampling, ~~two duplicates~~~~one duplicate~~ will be collected monthly.

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6.2.8.2 Rinse Blanks

A rinse blank is a sample of ~~distilled~~~~analyte free~~ water poured over or through a decontaminated field sampling equipment prior to the collection of environmental samples. Rinse blanks should be completed periodically to confirm that field sampling equipment is decontaminated. The field sampling technician will collect 1 rinse blank per month for laboratory analysis for the monthly Grassy sampling.

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7. Documentation and Records

9.0

Field data is at least as important as the analytical data received from the outside laboratory. Because field data includes an evaluation of the specific instantaneous conditions at the site, this information cannot be reproduced by a later trip to the site.

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~~The field data is often the first indication that there may be a concern with water quality at a given location. Therefore it is critical that any conditions observed are recorded in the field book.~~

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7.1. Field Notes / Field Log Book

~~Any necessary field notes will be recorded in a permanently bound, "waterproof" notebook. The year and area will be neatly written in permanent ink on the spine and front cover of each book. Xerox or electronic copies of all field notes will be maintained in the Victor Admin building and will be updated quarterly. This system ensures that the field data will be preserved in the event of fire or other unusual circumstance, as well as provide convenient access to this data by all department staff.~~

6.1

~~The field book contains compliance data, and therefore can be used in any compliance related proceedings (ie. inspections, NOAV hearings, trials, etc.).~~

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~~The field technician should strive to keep these notes suitably neat and well organized. Field notes shall be taken in pen with no erasures. Errors will be crossed out with a single line and corrected. The sampler will initial such corrections at the time they are made.~~

~~Field data recorded at each sampling site will include, at a minimum, pH, temperature, and sampling conditions (weather, etc.). Additional data that may be recorded, depending on the specific site are dissolved oxygen, conductivity, turbidity, flow, and depth to water. In many instances, careful recording of field observations has provided clues to questionable analytical~~

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~~results, thus saving considerable time and money. These observations may include water color, appearance, presence of floating matter or unusual amounts of suspended material, evidence of recent activity in the area or recent access by other persons, wildlife or stock, pumping rates (for monitor well samples), or any conditions that could conceivably impact water quality.~~

7.2.9.1 Field Data

The field data is often the first indication that there may be a concern with water quality at a given location. Therefore, it is critical that any conditions observed are recorded on the field sheets.

Field data recorded at each surface, and groundwater monitoring location shall be recorded on the applicable field sheet (surface, ~~or~~ groundwater or low flow sampling report) and will also be entered into the electronic data collection section for the Monitor Pro 5 Database Management System currently used by CC&V. Examples of the various field sheets are included in Appendix C. All field parameters collected and recorded on the completed field sheet will be entered into the electronic data entry for storage in our database. Each compliance sampling location has an associated electronic data entry form which is used to enter data.

Field data recorded at each sampling site will include, at a minimum, pH, temperature, and sampling conditions (weather, etc.). Additional data that may be recorded, depending on the specific site are dissolved oxygen, conductivity, turbidity, flow, and depth to water. In many instances, careful recording of field observations has provided clues to questionable analytical results, thus saving considerable time and money. These observations may include water color, appearance, presence of floating matter or unusual amounts of suspended material, evidence of recent activity in the area or recent access by other persons, wildlife or stock, pumping rates (for monitor well samples), or any conditions that could conceivably impact water quality. Pertinent observations during the sampling events will be recorded on the field sheets.

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~~7.3.~~ Calibration

9.2

All field instruments will be calibrated. The calibration and calibration check shall be documented on calibrations logs in the field book. Calibration data will not be used to alter any readings taken during the day. Calibration procedures shall follow the manufacturer's specifications.

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8.10.0 Collection and Preservation of Samples

The objective of sampling is to collect a representative sample that ensures the analytical results accurately represent the material being sampled. Following surface and groundwater sampling SOPs ensure that this is achieved. When alternative sampling methods are necessary due to unusual circumstances, the sampler will state plainly the nature of the modification in the field sheet log book.

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8.1.10.1 General Guidelines

A sampling event (day) should be scheduled to collect background samples first, and samples such as process solutions last, to avoid cross contamination of wells or streams. In general sampling should begin with locations that have the lowest

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concentrations of constituents and move to higher concentrations throughout the day.

Samples will be collected in new sample bottles of material consistent with the parameters to be analyzed. DO NOT touch the inside of sample vessel or cap or allow these surfaces to contact any material other than the sample media. Sample containers that are known, or suspected, to be contaminated will be discarded or clearly marked with an "X" or other designation to prevent their use. Holding times, minimum required sample bottles/volume, and necessary preservative types are contained within attachment I.

Table 108.1.1 below contains the required sample bottle, bottle volume, and preservative for the various analysis suites CC&V uses for our water monitoring program.

Table 10.1.1 - Sample Bottle Requirements by Analysis Suite

Table 8.1.1—Sample Bottle Requirements by Analysis Suite

Surface Water Analysis Suite
1 - 500 mL HDPE unpreserved unpreserved sample (grab sample) Black Label
1 - 250 mL HDPE unfiltered sample (grab sample) preserved with 1.25 mL H2SO4 (Sulfuric Acid, yellow label)
1 - 250 mL HDPE unfiltered sample (grab sample) preserved with 1.25 mL HNO3 (Nitric Acid, red label)
<u>1 - 250 mL HDPE filtered sample (grab sample) preserved with 1.25 mL HNO3 (Nitric Acid, red label)</u>
1 - 250 mL Amber HDPE unfiltered unpreserved sample (grab sample) preserved with 1.25 mL NaOH (Pre- preserved bottle or Sodium Hydroxide, green label)
<u>1 - Hexavalent Chromium Kit (pre-preserved) adjust pH as needed with NaOH (Sodium Hydroxide, Green Label)</u> <u>to achieve pH between 9.3 - 9.7. (All preservatives included in kit, along with instructions and pH strips).</u>

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1 - 250 mL HDPE filtered sample (grab sample) preserved with 1.25 mL Zn Acetate NaOH (Zinc Acetate NaOH, purple HNO₃ (Nitric Acid, red label))

Groundwater Analysis Suite

- 1 - 500 mL ~~HDPE~~ unpreserved sample (grab sample) Black Label
- 1 - 250 mL HDPE unfiltered sample (grab sample) preserved with 1.25 mL H₂SO₄ (Sulfuric Acid, yellow label)
- 1 - 250 mL HDPE unfiltered sample (grab sample) preserved with 1.25 mL HNO₃ (Nitric Acid, red label)
- 1 - 250 mL Amber HDPE unpreserved sample (grab sample) preserved with 1.25 mL NaOH (Sodium Hydroxide, green label)
- 1 - 250 mL HDPE filtered sample (grab sample) preserved with 1.25 mL HNO₃ (Nitric Acid, red label)

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Disposable latex surgical gloves will be worn during the collection and preservation of samples to minimize potential contamination of the sample, and to protect hands from preservatives and process water.

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When filling containers leave a small air space to allow for thermal expansion unless sampling for organics or dissolved oxygen which requires zero head space.

8.2.10.2 Field Book and Sample Field Sheets

A detailed record will be made at the time of collection of all pertinent information related to the sample. See Field Data Notes in the previous section (Section 9.07.1) for appropriate information. For locations requiring low-flow sampling protocols, a low flow sampling report ~~field sample sheet~~ will be ~~generated~~ used to document the process. Low Flow Sampling Reports are record stabilization criteria; these criteria include field parameters such as pH, conductivity, temperature, etc. An example low flow field sheet is included in Appendix C and the procedure is described in Appendix A as attachment I.

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8.3.10.3 Sample Identification

Gummed paper labels or tags will be filled out with waterproof ink at the time the sample is collected. The labels should contain the following information: date and

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time of sample collection, sample location, sample identification (ID#), name of sample collector, whether the sample was filtered, and type of preservative used. The labels must be attached to the appropriate sample bottle. In the absence of labels, write the above information directly on the sample bottle with a permanent marker.

Care must be exercised to ensure that the sample ID# is the same as the official designation for each sample location. Failure to use the same ID# as specified in the applicable permit may result in analytical results being questioned. Officially designated sample ID's for compliance sampling locations are specified in appendix G. Non-compliance samples to be collected shall be collected as necessary, with all necessary information being recorded within the sampler's field notebook. It is extremely important that sample identification and recorded notes be sufficient to identify precisely where the samples came from.

Duplicate and control samples will be identified with an ID# which will not bias the laboratory by indicating the origin of the sample. At CC&V duplicate samples are identified by increasing the numerical component of the monitor well identifier by 100, and offsetting the position nomenclature by positive 5 (Increasing the letter count by 5). For example the duplicate for monitoring well VIN-2A would be VIN-102F, the duplicate for monitoring well GVMW-22B would be GVMW-122G. It is also necessary that all pertinent field data for the duplicate sample be recorded as any sample, and that the sample ID is identified as a duplicate, and which well it is a duplicate for is identified.

Blank samples will be identified with an ID# which will not bias the laboratory by indicating the origin of the sample. At CC&V Blank samples are identified by using the prefix CCVB and then following it by the month numerical value and the day numerical value (two digit). For example the blank sample collected on March 25 would be identified as CCVB-0325. It is also necessary that field data be collected for the blank sample (pH, DO, EC, temp). The field data needs to be recorded on a field sample sheet.

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8.4.10.4 Sample Collection

Decontaminate all non-dedicated sampling equipment and meters, as appropriate, before and after use with deionized water (Type III reagent grade). Decontamination procedures include cleaning of equipment with a dilute phosphate free detergent solution (i.e. Alconox or Liquinox), followed by a fresh water rinse.

To assure an undiluted sample is collected, field cups, filter vessels, or other reusable equipment should be triple rinsed with sample solution if sufficient quantities are available.

Field readings will be measured from a separate container collected at the same time as the sample, and will not be taken from the actual sample bottle which will be analyzed. If a field sample was taken from a lined facility, it must be returned to a lined facility. Minimum field readings will be pH, conductivity, and temperature.

Sample collection from well, stream, pond, reservoir, & waste rock discussed below.

8.5.10.5 Well Sampling

To begin each sampling event, measure depth, to the nearest one-tenth of a foot, to static water level from the top of casing (TOC) with a water level indicator (Solinst). Rinse the level indicator with deionized water (type III reagent grade) before and after use.

A dry well will be recorded as "Dry at X feet" to assure that the Solinst did not hang up in the well.

Water standing in a well prior to sampling is not representative of in-situ ground water quality. Therefore, the stagnant water must be removed and replaced by fresh formation water. EPA protocol dictates that one to ten volumes of water standing in the well casing should be removed prior to collection of the sample. At CC&V the standard well sample collection method is the EPA Low Flow

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methodology (Low-Flow (minimal drawdown) Ground-water sampling procedures).

When sampling using EPA Low-Flow methodology, start the pump at low speed and slowly increase the speed until the discharge occurs. Record the pumping rate for future sampling and try to match the pumping rate used during previous sampling events. Otherwise adjust the speed until there is little to no water level drawdown. If the minimal drawdown exceeds 0.3 feet, but remains stable, continue purging. Monitor and record the water level and pumping rate at least every five minutes during purging. Record any pumping rate adjustments (both time and flow rate). Pumping rates should, as needed, be reduced to the minimum capabilities of the pump to ensure stabilization of the water level. Adjustments are best made in the first fifteen minutes of pumping in order to help minimize purging time. During pump start-up, drawdown may exceed the 0.3 feet target and then "recover" somewhat as pump flow adjustments are made. Purge volume calculations should utilize stabilized drawdown value, not the initial drawdown. If the initial water level is above the top of the screen do not allow the water level to fall into the well screen. The final purge volume must be greater than the stabilized drawdown volume plus the pump's tubing volume. If the drawdown has exceeded 0.3 feet and stabilizes, calculate the volume of water between the initial water level and the stabilized water level. Add the volume of the water which occupies the pump's tubing to this calculation. This combined volume of water needs to be purged from the well after the water level has stabilized before samples are collected. During purging monitor field parameters (pH, temperature, & conductivity). Purging is considered complete and sampling may begin when all the indicator field parameters (below) have stabilized. Achievement of turbidity levels of less than 5 NTU, and stable drawdowns of less than 0.3 feet, while desirable, are not mandatory.

The EPA sampling method is designed for ideal conditions and wells. Wells and conditions at CC&V can be variable and it should always be the goal of this sampling program to collect meaningful data. If deviations are made from the standard procedure, they will be documented and done so in the interest of collecting data that is useful.

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~~Some~~However, some wells yield such low volumes of water that this protocol cannot be followed. Wells containing less than 5 feet of water column will be considered insufficient and no samples will be collected. When sampling a low yield well evacuate the well to dryness once. Within 24 hours of this purge, if the well has recovered at least 90% of the original volume collect, preserve, and handle the sample(s) according to normal procedures. This procedure described mimics that described by the United States Geologic Survey's National Field Manual for the Collection of Water-Quality Data.

~~Another exception would be dedicated low flow pumping technologies (see section 9.1).~~

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Standard/Well-Volume Method may also be utilized forWhen sampling monitoring wells, for this methoda high yield well, three casing volumes will be evacuated prior to sampling. Measure the stabilization criteria outlined belowpH, temperature, and conductivity after each well volume is evacuated (i.e., ~~ie~~, if the well volume is 5 gallons, take measurements after evacuation of 5, 10, and 15 gallons). On~~In~~ the field formbook, record the volume of water evacuated, all stabilization parameters~~the pH, temperature, conductivity~~, and time that the measurements were made. After three well volumes have been purged (appendix E, well evacuation calculation) check the last two sets of measurements to determine if the field parameters have stabilized. If the field readings have not stabilized purge another well volume and take field measurements. Repeat until stabilized.

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Stabilization criteria are achieved~~met~~ when the following parameters are met over three consecutive readings.

- ~~±~~ **Temperature** - $\pm 3\%$
- $\pm 1.0^{\circ}\text{C}$, **Specific Conductivity** - $\pm 3\%$
- $\%$, **pH** - ± 0.1
- **Dissolved Oxygen** - (10% for values greater than 0.5 mg/L, if three Dissolved Oxygen values are less than 0.5 mg/L, consider the values as stabilized)

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- **Turbidity** - (10% for values greater than 5 NTU; if three Turbidity values are less than 5 NTU, consider the values as stabilized)
- **Oxidation/Reduction Potential** - ± 10 millivolts

If the field values indicate stable conditions, collect, preserve, and handle the samples according to the procedures outlined in this document. An example of a well purging calculation is included in Appendix E.

8.6.10.6 Stream Sampling

Sampling results will vary with depth, stream flow, and distance from shore. When rinsing the field cup or bottles, discard the rinsate downstream from the sampling point. A grab sample should be collected at mid-depth from the middle of the stream, in an actively flowing section of the stream. In shallow streams, care should be taken not to disturb the bottom and put sediments into suspension, as these will affect the analytical results. If a stream has no visible flow, it will be recorded as dry or stagnant and will not be sampled. If a stream sampling location has visible water, but not enough water to allow collection of a representative sample, it will be recorded as "too low to sample". Sampling data will be recorded on~~An estimate of the flow rate of water at each stream sampling logs. location should be recorded in the field book, along with the general appearance of the water (turbidity, color, etc.).~~

8.7.10.7 Filtering Samples

A ground water sample to be analyzed for dissolved metals must be passed through a 0.45 micron membrane filter prior to preservation. For the determination of total metals, the sample is not filtered. As a general guideline, ground water (wells) samples requiring a metals analysis should be filtered and analyzed for dissolved metals, while surface water samples requiring a metals analysis should be unfiltered and analyzed for total metals. Samples requiring analysis for organics should not be filtered. Specify on the Newmont chain of custody whether or not the samples have been filtered.

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8.8.10.8 Sample Preservation and Storage

Sample preservation is intended to retard breakdown of the constituents within the sample. Preservation methods include pH control, chemical control, temperature control, and protection from light. Common sample preservation measures include the following:

To avoid changes in the concentration or physical state of the constituent to be analyzed, preserve accordingly, and pack samples in ~~ice~~ in the field and when shipping. Samples should be stored in a locked refrigerator and shipped to the laboratory as soon as possible. As samples are transferred to the storage refrigerator, a double check that the lids are securely tightened is conducted.

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8.9.10.9 Sample Handling and Custody

The contracted laboratory will provide necessary coolers, sample bottles, chain-of-custody (COC) forms and shipping labels. After sample collection, samples will be stored in a refrigerator and shipped to the laboratory as soon as possible and within allowable holding times. Samples will be cooled to temperature acceptable to the contracted labratory $\leq 4^{\circ}\pm 2^{\circ}\text{C}$. Sample containers will be packed to prevent breakage or contamination during shipment.

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8.10.10.10 Chain of Custody Procedures

Chain of custody procedures will allow for the tracking of individual samples from the time of collection through laboratory analysis. All records relating to chain of custody documentation are to be made in ink. If errors are made on any of these documents, corrections are to be made by crossing a single line through the mistake and entering the correct information. All corrections are to be initialed and dated by the individual making the error, if possible, or by the investigator. All paperwork completed in the course of collecting and shipping samples must be correct, accurate, and defensible in a court of law.

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The complete COC will accompany the sample from the site, through delivery to the contracted laboratory. To ensure complete documentation of sample

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custody, field personnel and laboratory personnel will sign and date the COC upon shipment and receipt.

~~8.11.~~**10.11** ~~Sampling Field Sheets~~**Log Book**

The ~~sampling logs~~**field log book** contains the first record in the chain of custody of the sample. It is previously discussed in Section ~~94.0~~ (Field Data) of this document.

~~8.12.~~**10.12** ~~Chain of Custody Record~~

A chain of custody record will be completed and must accompany each sample or each cooler of samples (see Appendix A). The record will include the following information: specific area/permit name, sample identification, sample type (well, grab, soil, other), preservative(s) used, whether the sample was filtered, type of analysis(es) required, number of sample containers, signatures of persons involved in the chain of possession, inclusive dates of possession, a unique ~~sample ID, NMC identification number which includes the area (NA = North Area, SA = South Area, RN = Rain Mine, EM = Emigrant Mine, WA = Waste, HO= Hollister) and date, cooler number,~~ and method of shipment. If an analysis is to be rushed, state this on the Chain of Custody. One copy of the chain of custody is kept by Newmont. The laboratory keeps one copy for its records, and returns a copy to Newmont with the analysis reports.

~~8.13.~~**10.13** ~~Shipping Papers~~

A shipping label is attached to the top of cooler or shipping container along with a laboratory's address label. The cooler or shipping container must be secured with shipping fasteners or packing tape to prevent opening during transportation.

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~~8.14.~~10.14 Delivery to Laboratory

Planning is required to ship the samples, so the lab personnel are available to receive them, especially if shipping over the weekend. If a rush analysis is requested, notify the laboratory ahead of time. To ship samples, fill out a shipping paper for each cooler (see Appendix C) and deliver the coolers and shipping papers to the shipper. Make a copy of the shipping paper, COC and file in the Chain of Custody folder (see Tracking System section below). **Note:** For water quality samples there must be enough ice in the cooler to ensure the sample temperature is maintained at a temperature that is acceptable to the laboratory, $\leq 4^{\circ} \pm 2^{\circ} \text{C}$.

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~~8.15.~~10.15 Data Receipt and Review

Analysis reports are received with the invoices. The data should be reviewed relative to the applicable standards, and invoices filed appropriately. Should any analytical results deviate from expectation or exceed the applicable standards the coordinator should be informed.

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~~9.11.0~~ Well Sampling Equipment

If possible, dedicated sampling devices should be used in water quality monitoring. When non-dedicated equipment is used, equipment will be decontaminated using the procedures detailed above, before and after use to avoid cross contamination of wells. Equipment should be used according to manufacturers' guidelines. A brief summary of guidelines for various types of equipment used by CC&V follows:

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~~9.11.1~~ Dedicated Electric Powered Sampling Pump

When possible, use the dedicated electric powered sampling pump to purge sampling wells. If the electrical connection coming from the dedicated sampling pump is a 240 volt connection, connect directly to the 240 volt input on the

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generator. If a four prong electrical connection comes from the dedicated sampling pump, connect the four prong connection from the sampling pump to the four prong connection on the control box. Then connect the 240 volt connection on the flow control box with the 240 volt input on the generator. Once the electrical connections are made pull out the choke on the generator, ensure the gas valve is open and start the generator. Once the generator is started push in the choke, if the control box is connected, adjust the flow rate using the nob on the control box. The optimum flow rate should allow at least three well volumes to be purged without purging the well dry.

After sampling turn off the generator and close the gas valve. Disconnect the electrical connections from the sample pump, control box, and generator. Ensure the sampling pump's electric connection and wiring is inside the well casing and secured below the well cap.

9.2.11.2 Field Deployable Submersible Pump

When a dedicated electric powered sampling pump is not available for use at the monitoring well to be sampled, a field deployable submersible pump is to be used. Currently CC&V uses a Geotech Environmetnal SS Geosub or a Grunfos Redi-Flo 2 portable submersible pump for sample collection. This submersible pump has an external controller to manage the pump. To use the pump follow the instructions and training provided to you, if you have not received the necessary or adequate instruction contact your supervisor to receive the necessary training. Insure that all electrical connections are solid, and that the necessary grounding is in place. Operate the necessary power equipment (generator or inverter) to power the controller to collect the sample.

After sampling turn off the generator and close the gas valve (if used). Disconnect the electrical connections from the sample pump, control box, and generator.

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9.3.11.3 Bailer

The hand bailer is a cylindrical tube, constructed of either plastic or metal, with a ball valve at the bottom. The ball valve allows water into the tube while the bailer is descending and closes, thus trapping water, when the bailer is ascending.

The bailer is to be decontaminated as described above, before and after use. The bailer is lowered into the well with a nylon rope, or a stainless-steel mesh rope, which is decontaminated after each use. Care must be taken to ensure that the rope does not come into contact with the ground while sampling, to avoid any possible contamination. A disposable bailer must be used and then discarded when sampling for VOCs. Otherwise, it may be decontaminated and reused.

The line should be securely affixed to the bailer, and the opposite end should be securely affixed to something to prevent losing the bailer

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11.4 Water Quality Meters

CC&V utilizes primarily 3 types of water quality meters to collect field parameters. These meters are listed below along with the associated appendix containing calibration information:

- Aqua Troll 600 (preferred for all samples; Appendix K)
- YSI Pro (Appendix G)
- Myron Ultrameter (Appendix K)

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APPENDIX A
CHAIN OF CUSTODY RECORD

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FOR SVL USE ONLY
SVL Work Order #

Temperature on Receipt:

Table 1 - Matrix Type
1 = Surface Water, 2 = Ground Water
3 = Soil, 4 = Sediment, 5 = Rock, 6 = Flammable, 7 = Oil
8 = Waste, 9 = Other:

CHAIN OF CUSTODY RECORD
SVL Analytical, Inc. • One Government Gulch • Killeeg, ID 83837 • (208) 784-1258 • FAX: (208) 785-0891

Page _____ of _____

Report to Company:
Contact:
Address:
Phone Number:
FAX Number:
Email:

Invoice Sent To:
Contact:
Address:
Phone Number:
FAX Number:
POB:

Project Name:
Sample's Signature:

Indicate State of sample origination:

Sample ID	Collection	Date	Time	Matrix Type (From Table 1)	No. of Containers	Unfiltered	HNO ₃ Filtered	HNO ₃ Unfiltered	HCl	H ₂ SO ₄	NaOH	Other (Specify)	Preservative(s)	Misc.
1 and 1 2 and 2 3 and 3 4 and 4 5 and 5 6 and 6 7 and 7 8 and 8 9 and 9 10														
Please take care to distinguish between: 1 and 1 2 and 2 3 and 3 4 and 4 5 and 5 6 and 6 7 and 7 8 and 8 9 and 9 10														
Thanks!														

Analyses Required

Rush Instructions (Days)

Comments

Return ☐ Deposit ☐ Store (30 Days) ☐

Requisitioned by: _____ Date: _____ Time: _____

Received by: _____ Date: _____ Time: _____

* Sample Reject: ☐ ☐ ☐

SVL-COC-0717

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APPENDIX B
Field Sampling Bottle Requirements and Holding Times

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Holding Times and Containers for Water/Aqueous Samples					
General Chemistry / Metals	Method	Holding Time (days)	Min Volume (ml)	Container	Preservation
Alkalinity	SM 2220 B	14	100	250-mL HDPE	4°C
Biochemical Oxygen Demand (BOD)	SM 5210 B	48 hours	300	1-L HDPE	4°C
Bromide	EPA 300.0	28	20	125-mL HDPE	4°C
Carbon Dioxide	SM 4500-CO ₂ D	24 hours	250	250-mL amber glass	4°C (no headspace)
Carbon Dioxide	RSK 175(M)	7	40	2 + 40-mL VOA vials	4°C (no headspace)
Chemical Oxygen Demand (COD)	SM 5220 D	28	250	250-mL glass	H ₂ SO ₄ & 4°C
Chloride	EPA 300.0 / SM 4500-Cl ⁻ G	28	50	125-mL HDPE	4°C
Chlorine, Total Residual	SM 4500-Cl ⁻ F	15 minutes	100	500-mL HDPE	4°C
Chromium VI (Hexavalent Chromium)	EPA 218.6 / 7196A / 7199	24 hours	200	250-mL HDPE	4°C
Cyanide, Amenable	SM 4500-CN ⁻ G	14	500	1-L HDPE	NaOH & 4°C
Cyanide, Total	SM4500-CN ⁻ G/E	14	500	1-L HDPE	NaOH & 4°C
Dissolved Oxygen	SM 4500-O ₂ G	15 minutes	300	500-mL amber glass	4°C (no headspace)
Ferrous Iron	SM 3500-Fe B	24 hours	50	250-mL amber glass	4°C (no headspace)
Ferrous Iron	SM 3500-Fe B	24 hours	50	250-mL amber glass	HCl & 4°C (no headspace; field filtered)
Fluoride	SM 4500-F ⁻ C	28	100	125-mL HDPE	4°C
Formaldehyde	ASTM D6303-08	24 hours	150	500-mL amber glass	4°C
Hardness, Total / Calcium	SM 2340 C / SM 3500-Ca B	180	100	250-mL HDPE	HNO ₃
Hexane Ext. Material (HEMISOT-HEM)	EPA 1664A	28	1000	1-L amber glass	H ₂ SO ₄ & 4°C
Ignitability (Flashpoint)	EPA 1010A	14	250	250-mL HDPE	4°C
Mercurians	LACSD 266	48 hours	50	125-mL HDPE	4°C
Mercury	EPA 7470A / 245.1	28	100	250-mL HDPE	HNO ₃
Metals (ICP)	EPA 8010B / 200.7	180	100	250-mL HDPE	HNO ₃
Metals (ICP/MS)	EPA 8020 / 200.8	180	100	250-mL HDPE	Ultra HNO ₃
Nitrogen, Ammonia (NH ₃)	SM 4500-NH ₃ B/C	28	500	1-L amber glass	H ₂ SO ₄ & 4°C
Nitrogen, Nitrate (NO ₃)	EPA 300.0 / SM 4500-NO ₃ E	48 hours	50	125-mL HDPE	4°C
Nitrogen, Nitrite (NO ₂)	EPA 300.0 / SM 4500-NO ₂ B	48 hours	50	125-mL HDPE	4°C
Nitrogen, Nitrate+Nitrite (NO ₃ +NO ₂)	SM 4500-NO ₃ E / SM 4500-NO ₂ B	28	50	125-mL HDPE	H ₂ SO ₄ & 4°C
Nitrogen, Total Kjeldahl (TKN)	SM 4500-N _T B	28	500	1-L amber glass	H ₂ SO ₄ & 4°C
Nitrogen, Total	TKN / NO ₃ + NO ₂	28	500	1-L amber glass	H ₂ SO ₄ & 4°C
Nitrogen, Total Inorganic	NH ₃ / NO ₃ + NO ₂	28	500	1-L amber glass	H ₂ SO ₄ & 4°C
Nitrogen, Total Organic	TKN - NH ₃	28	1000	1-L amber glass	H ₂ SO ₄ & 4°C
Oil and Grease	SM 5520 B	28	1000	1-L amber glass	H ₂ SO ₄ & 4°C
Oil and Grease	EPA 413.2	28	500	500-mL amber glass	H ₂ SO ₄ & 4°C
Organic Lead	DHS LUTT	7	100	500-mL amber glass	4°C
Perchlorate	EPA 314.0 / 331.0(M)	28	50	125-mL / 100-mL sterile HDPE	4°C
pH	SM 4500-4 ⁺ B	15 minutes	50	125-mL HDPE	4°C
Phenolics, Total	EPA 420.1	28	200	500-mL amber glass	H ₂ SO ₄ & 4°C
Phosphate, Ortho	EPA 300.0 / SM4500-P B/E	48 hours	50	125-mL HDPE	4°C
Phosphate, Total	SM 4500-P B/E	28	100	250-mL glass	H ₂ SO ₄ & 4°C
Phosphorus, Dissolved	SM 4500-P B/E	28	100	250-mL glass	4°C
Phosphorus, Total	SM 4500-P B/E	28	100	250-mL glass	H ₂ SO ₄ & 4°C
Redox Potential	ASTM D-1498	24 hours	50	125-mL HDPE	4°C
Salinity	SM 5520 B	28	100	125-mL HDPE	4°C
Solids, Total Dissolved (TDS)	SM 2540 C	7	1000	1-L HDPE	4°C
Solids, Total Suspended (TSS)	SM 2540 D	7	1000	1-L HDPE	4°C
Solids, Total (TS)	SM 2540 B	7	200	500-mL HDPE	4°C
Solids, Volatile (V _S)	SM 2540 E / EPA 160.4	7	200	500-mL HDPE	4°C
Solids, Settleable (SS)	SM 2540 F	48 hours	1000	1-L HDPE	4°C
Solids, Volatile Suspended (V _{SS})	SM 2540 D / EPA 160.4	7	1000	1-L HDPE	4°C
Specific Conductance	SM 2510 B	28	50	125-mL HDPE	4°C
Sulfate	EPA 300.0 / ASTM D516-02	28	50	125-mL HDPE	4°C
Sulfide, Soluble	SM 4500-S ²⁻ D	15 minutes	50	125-mL HDPE	4°C
Sulfide, Total	SM 4500-S ²⁻ D	7	50	125-mL HDPE	ZnAc ₂ & NaOH & 4°C
Surfactants (MBAS)	SM 5540 C	48 hours	200	500-mL HDPE	4°C
Thiosulfate	LACSD 253A	48 hours	200	500-mL HDPE	4°C
Total Organic Carbon (TOC)	SM 5210 D	28	150	250-mL glass	H ₂ SO ₄ & 4°C
Turbidity	SM 2100 B	48 hours	100	125-mL HDPE	4°C
96-Hour Aquatic Toxicity, Haz Waste	CA Dept. Fish & Game	28	100	250-mL HDPE	4°C
Volatiles / Semi-Volatile Organics					
EDB/DBCP	EPA 804.1	14	40	3 + 40-mL VOA vials	Na ₂ S ₂ O ₅ & 4°C
Ethanol (low level)	EPA 824.2(M) / SM 18200(B)(M) SIM	14	40	3 + 40-mL VOA vials	HCl & 4°C (no headspace)
EPH	EPA 8015(B)(M)	14	500	500-mL amber glass	H ₂ SO ₄ & 4°C
Herbicides, Chlorinated	EPA 8151A	7	1000	1-L amber glass	4°C
Methane in Water	RSK 175(M)	14	40	2 + 40-mL VOA vials	HCl & 4°C (no headspace)
Methane / Ethanol	EPA 8015(B)	14	40	2 + 40-mL VOA vials	4°C (no headspace)
NDA	EPA 1605(C)(M)	7	1000	1-L amber glass	4°C
Organonitros	Krueger et al	7	1000	1-L amber glass	4°C
PCBs	EPA 8082 / 606	7	1000	1-L amber glass	4°C
Pesticides, Organochlorine	EPA 9081A / 608	7	1000	1-L amber glass	4°C
Pesticides, Organophosphorus	EPA 8141(B)	7	1000	1-L amber glass	4°C
SVOCs (BNA)	EPA 8270C / 625	7	1000	1-L amber glass	4°C
TPH(CC) / TPH(G) / DRO	EPA 8015(B)(M) / 8015(B) / 8015(B)	7	500	500-mL amber glass	4°C
TPH(G) / GRO / BTEX / MTBE	EPA 8015(B)(M) / 8015(B) / 8021(B) / 602	14	40	3 + 40-mL VOA vials	HCl & 4°C (no headspace)
TPRH	EPA 410.1	28	500	500-mL amber glass	H ₂ SO ₄ & 4°C
VOCs / TPHH	EPA 8260B / 624 / 1-LIFT / GC/MS	14	40	3 + 40-mL VOA vials	HCl & 4°C (no headspace)
VOCs (drinking water)	EPA 624.2	14	40	3 + 40-mL VOA vials	Ascorbic Acid / HCl & 4°C (no headspace)
Volatiles Fatty Acids (Organic Acids)	HPLC/UV	28	40	2 + 40-mL VOA vials	H ₃ PO ₄ & 4°C (no headspace)
VPH	EPA 8260B	14	40	3 + 40-mL VOA vials	HCl & 4°C (no headspace)
1,2,3-TCDF	SRL-624M/TCF	14	40	3 + 40-mL VOA vials	HCl & 4°C (no headspace)
1,4-Dioxane	GC/MS Isotope Dilution	7	1000	1-L amber glass	4°C
* days for extraction; 40 days after extraction for analysis.					
eurofins		Laboratory Location: 7440 Lincoln Way Garden Grove, CA 92641-1427		For information, please contact Sales Department at (714) 895-6494. fax: (714) 894-7551 or e-mail: us25_sales@eurofinsus.com	
Calscience				2014-08-07 Revision	

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Holding Times and Containers for Soil/Solid Samples					
General Chemistry / Metals	Method	Holding Time (days)	Minimum Mass (g)	Container	Preservation
Alkalinity	SM 2220 B	14	20	4-oz glass jar w/effen lid	4°C
Biochemical Oxygen Demand (BOD)	SM 5210 B(M)	48 hours	30	4-oz glass jar w/effen lid	4°C
Bromide	EPA 300.0(M)	28	10	4-oz glass jar w/effen lid	4°C
Chemical Oxygen Demand (COD)	SM 5220 D(M)	28	10	4-oz glass jar w/effen lid	4°C
Chloride	EPA 300.0(M)	28	10	4-oz glass jar w/effen lid	4°C
Chromium VI (Hexavalent Chromium)	EPA 7196A / 7199 / 3060A	30	10	4-oz glass jar w/effen lid	4°C
Chromium VI (Hexavalent Chromium)	EPA 7199 / 3060A	30	10	4-oz glass jar w/effen lid	4°C
Cyanide, Amenable	EPA 9010C / 9014	14	20	4-oz glass jar w/effen lid	4°C
Cyanide, Reactive	SW 846 Ch. 7	14	20	4-oz glass jar w/effen lid	4°C
Cyanide, Total	EPA 9010C / 9014	14	10	4-oz glass jar w/effen lid	4°C
Ferrous Iron	SM 3500-Fe B(M)	24 hours	10	4-oz glass jar w/effen lid	4°C
Fluoride	SM 4500-F ⁻ C(M)	28	20	4-oz glass jar w/effen lid	4°C
Hexane Ext. Material (HEMSTG-HEM)	EPA 1664(M)	28	30	4-oz glass jar w/effen lid	4°C
Ignitability	EPA 1000	14	100	4-oz glass jar w/effen lid	4°C
Mercury	EPA 7471A	28	1	4-oz glass jar w/effen lid	None
Metals	EPA 6010B / 6020	180	2	4-oz glass jar w/effen lid	None
Moisture Content	ASTM D2216	10	20	4-oz glass jar w/effen lid	4°C
Nitrogen, Ammonia	SM 4500-NH ₄ B(C)(M)	28	10	4-oz glass jar w/effen lid	4°C
Nitrogen, Nitrate	EPA 300.0(M)	7	10	4-oz glass jar w/effen lid	4°C
Nitrogen, Nitrite	EPA 300.0(M)	7	10	4-oz glass jar w/effen lid	4°C
Nitrogen, Nitrate+Nitrite (NO ₃ +NO ₂)	SM 4500-NO ₃ E(M) / SM 4500-NO ₂ B(M)	7	10	4-oz glass jar w/effen lid	4°C
Nitrogen, Organic	SM 4500-NH ₄ / 4500-NH ₄ B	28	10	4-oz glass jar w/effen lid	4°C
Nitrogen, Total Kjeldahl (TKN)	SM 4500-NH ₄ B(M)	28	10	4-oz glass jar w/effen lid	4°C
Nitrogen, Total	TKN / NO ₃ + NO ₂	7	30	4-oz glass jar w/effen lid	4°C
Oil and Grease	SM 5620 B(M)	28	30	4-oz glass jar w/effen lid	4°C
Organic Lead	DHS LUFT	14	10	4-oz glass jar w/effen lid	4°C
Perchlorate	EPA 314.0(M) / 6050	28	20	4-oz glass jar w/effen lid	4°C
pH	EPA 9240D	ASAP (24 hours)	20	4-oz glass jar w/effen lid	4°C
Phenolics, Total	EPA 8065	28	20	4-oz glass jar w/effen lid	4°C
Phosphate, Ortho	EPA 300.0(M)	7	10	4-oz glass jar w/effen lid	4°C
Phosphate, Total	SM 4500-P BIE(M)	28	20	4-oz glass jar w/effen lid	4°C
Phosphorus, Total	SM 4500-P BIE(M)	28	20	4-oz glass jar w/effen lid	4°C
Specific Conductance	EPA 9050A	28	20	4-oz glass jar w/effen lid	4°C
Sulfate	EPA 300.0(M) / 9038	28	20	4-oz glass jar w/effen lid	4°C
Sulfide, Reactive	SW 846 Ch. 7	7	20	4-oz glass jar w/effen lid	4°C
Sulfide, Total	SM 4500-S ²⁻ D	7	20	4-oz glass jar w/effen lid	4°C
Surfactants (SBAS)	SM 5640 C(M)	48 hours	20	4-oz glass jar w/effen lid	4°C
Total Organic Carbon (TOC)	EPA 9060A	28	2	4-oz glass jar w/effen lid	4°C
24-Hour Aquatic Toxicity, Haz Waste	CA Dept. Fish & Game	28	100	4-oz glass jar w/effen lid	4°C
Volatile / Semi-Volatile Organics					
Analysis	Method	Holding Time (days)	Minimum Mass (g)	Container	Preservation
APV	EPA 8015B(M)	14*	10	4-oz glass jar w/effen lid	4°C
Herbicides, Chlorinated	EPA 8151A	14*	50	4-oz glass jar w/effen lid	4°C
Methanol / Ethanol	EPA 8015B	14	50	4-oz glass jar w/effen lid	4°C
Oil and Grease	EPA 413.2(M)	28	5	4-oz glass jar w/effen lid	4°C
Organotins	Krone et al	14*	20	4-oz glass jar w/effen lid	4°C
PAHs	EPA 8210	14*	20	4-oz glass jar w/effen lid	4°C
PCBs	EPA 8082	14*	20	4-oz glass jar w/effen lid	4°C
Pesticides, Organochlorine	EPA 8081A	14*	20	4-oz glass jar w/effen lid	4°C
Pesticides, Organophosphorus	EPA 8141B	7*	20	4-oz glass jar w/effen lid	4°C
SVOCs (BNA)	EPA 8210C	14*	20	4-oz glass jar w/effen lid	4°C
TPH(G) / TPH(G) / DRO	EPA 8015B(M) / 8015B / 8015B	14*	10	4-oz glass jar w/effen lid	4°C
TPH(G) / GRO / BTEX / MTBE	EPA 8015B(M) / 8015B / 8021B	14	5	4-oz glass jar w/effen lid	4°C
TPH(G) / GRO (Bg EnCore Sampler)	EPA 5035 / 8015B(M) / 8015B	48 hours**	2/sample	2 EnCores	4°C
TPH(G) / GRO (Bg TerraCore Sampler)	EPA 5035 / 8015B(M) / 8015B	14	2/sample	2 TerraCores	4°C
TPH	EPA 8151.1(M)	14	5	4-oz glass jar w/effen lid	4°C
VOCs / TPH	EPA 8260B / LUFT GC/MS	14	10	4-oz glass jar w/effen lid	4°C
VOCs (Bg EnCore Sampler)	EPA 5035 / 8260B	48 hours**	3/sample	3 EnCores	4°C
VOCs (Bg TerraCore Sampler)	EPA 5035 / 8260B	14	3/sample	3 TerraCores	4°C
VPH	EPA 8260B	14	5	4-oz glass jar w/effen lid	4°C
VPH (Bg EnCore Sampler)	EPA 5035 / 8260B	48 hours**	3/sample	3 EnCores	4°C
VPH (Bg TerraCore Sampler)	EPA 5035 / 8260B	14	3/sample	3 TerraCores	4°C
* days for extraction; 48 days after extraction for analysis.					
** hours for extraction; 14 days for analysis.					
STLC / TCLP or SPLP					
Analysis	Method	Holding Time (days)	Minimum Mass (g)	Method Ext. After	Holding Time After Ext. (days)
Mercury	COR T22.11.5.A-II / EPA 1311/1312	28	50 / 100	NA	28
Metals	COR T22.11.5.A-II / EPA 1311/1312	180	50 / 100	NA	180
SVOCs	COR T22.11.5.A-II / EPA 1311/1312	14	50 / 100	NA	48
TPH(G) / GRO	COR T22.11.5.A-II / EPA 1311/1312	14	50 / 100	7	48
TPH(G) / GRO	COR T22.11.5.A-II / EPA 1311/1312	14	50 / 25	NA	7
VOCs	COR T22.11.5.A-II / EPA 1311/1312	14	50 / 25	NA	7
Holding Times and Containers for Air/Vapor Samples					
Analysis	Method	Holding Time (days)	Minimum Volume (L)	Container	Preservation
Fixed Gases	ASTM D1948	3 / 30	1	Tedlar Bag / Summa Canister	Keep out of sunlight
Hydrocarbon Speciation	ASTM D2000	3 / 30	1	Tedlar Bag / Summa Canister	Keep out of sunlight
Hydrogen Sulfide (H ₂ S)	GC/FPD	24 hours	1	Tedlar Bag / Silica Canister	Keep out of sunlight
Landfill Gases (NMOCs)	SCAQMD 25.1(M)	3 / 30	1	Tedlar Bag / Summa Canister	Keep out of sunlight
TPH(G)	TO-3(M)	3 / 30	1	Tedlar Bag / Summa Canister	Keep out of sunlight
VOCs	EPA TO-14A/TO-15	3 / 30	1	Tedlar Bag / Summa Canister	Keep out of sunlight
<div>  <div> <div>Laboratory Location:</div> <div>7440 Lincoln Way</div> <div>Garden Grove, CA 92641-1427</div> </div> <div> <div>For information, please contact:</div> <div>John Thompson at (714) 895-6454,</div> <div>fax: (714) 894-7501 or e-mail: us30_sales@eurofinsus.com</div> </div> </div>					
2014-08-07 Revision					

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APPENDIX C
Example Field Sample Record Sheets

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Groundwater Sampling Log

Location : _____ Date: _____

Technician: _____ Quarter: _____

Static Water Level (DTW): _____ Well ID: _____

Is well Drv? _____ Well Depth (TD): _____

If so Drv at: _____ feet

[illegible]

Sample Method: _____ Rate (gpm): _____ Time Start: _____ Time End: _____
** Flow rate at stabilization (during sample collection)*

Final Parameters	Stabilization Guidance	Met?	Comments
pH	10.1	Y / N	
Conductivity	3%	Y / N	
Temp (deg C)	3%	Y / N	
Dissolved Oxygen	10%	Y / N	
Turbidity	10%	Y / N	
Oxidation/Reduction Potential	±10	Y / N	
DTW Stabilized	feet	Y / N	
Final H2O level	feet		

If Low Flow Method: Drawdown greater than 0.33 ft? Y / N If yes, required pump vol (gal): _____ Actual vol. pumped (gal) _____
 * See Field Volume Guide following stabilization

O/G visible:	Y / N	Turbid?	Y / N
Equipment Decontaminated:	Y / N		

Decontamination procedure used:

Weather: _____
Signature: _____

<p>Volume Calculations:</p> <p>For 2" Diameter Well (gal): $V(gal) = 0.1632 \cdot h(ft)$ For 4" Diameter Well (gal): $V(gal) = 0.6528 \cdot h(ft)$</p> <p>Other Diameter Well & Tubing Vol (gal): $V(gal) = 0.1632 \cdot (r(in))^2 \cdot h(ft)$</p> <p>Water Column Calculation: $h(ft) = Total\ Depth(TD)(ft) - Depth\ to\ Water(DTW)(ft)$</p> <p>Well Volume Purge Method: Three Well Volumes = 3*V</p>	
<p>Conversions:</p> <p>1ft³ = 7.48 gal</p> <p>1gal = 3.785 L</p>	<p>Show Calculations:</p>

Newmont Mining Co

Cripple Creek & Victor Gold Mining Co

Surface Water Sampling Log

Location : _____

Date: _____

Technician: _____

Quarter: _____

Time	pH (S.U.)	Cond. (uS/cm)	Temp. (°C)	Notes

Sample Method: _____

Oil/Gas visible [Y / N]

Turbid [Y / N]

Clear [Y/N]

Weather: _____

Signature: _____

Comments:

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	119	120	121	122	123	124	125	126	127	128	129	130	131	132	133	134	135	136	137	138	139	140	141	142	143	144	145	146	147	148	149	150	151	152	153	154	155	156	157	158	159	160	161	162	163	164	165	166	167	168	169	170	171	172	173	174	175	176	177	178	179	180	181	182	183	184	185	186	187	188	189	190	191	192	193	194	195	196	197	198	199	200	201	202	203	204	205	206	207	208	209	210	211	212	213	214	215	216	217	218	219	220	221	222	223	224	225	226	227	228	229	230	231	232	233	234	235	236	237	238	239	240	241	242	243	244	245	246	247	248	249	250	251	252	253	254	255	256	257	258	259	260	261	262	263	264	265	266	267	268	269	270	271	272	273	274	275	276	277	278	279	280	281	282	283	284	285	286	287	288	289	290	291	292	293	294	295	296	297	298	299	300	301	302	303	304	305	306	307	308	309	310	311	312	313	314	315	316	317	318	319	320	321	322	323	324	325	326	327	328	329	330	331	332	333	334	335	336	337	338	339	340	341	342	343	344	345	346	347	348	349	350	351	352	353	354	355	356	357	358	359	360	361	362	363	364	365	366	367	368	369	370	371	372	373	374	375	376	377	378	379	380	381	382	383	384	385	386	387	388	389	390	391	392	393	394	395	396	397	398	399	400	401	402	403	404	405	406	407	408	409	410	411	412	413	414	415	416	417	418	419	420	421	422	423	424	425	426	427	428	429	430	431	432	433	434	435	436	437	438	439	440	441	442	443	444	445	446	447	448	449	450	451	452	453	454	455	456	457	458	459	460	461	462	463	464	465	466	467	468	469	470	471	472	473	474	475	476	477	478	479	480	481	482	483	484	485	486	487	488	489	490	491	492	493	494	495	496	497	498	499	500	501	502	503	504	505	506	507	508	509	510	511	512	513	514	515	516	517	518	519	520	521	522	523	52
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Newmont Mining Co
Cripple Creek & Victor Gold Mining Co

Groundwater Sampling Log

Location : _____ Date: _____

Technician: _____ Quarter: _____

Static Water Level: _____ Well ID: _____

Is well Dry? _____ If so Dry at: _____ feet

[illegible]

Sample Method: _____ Rate (gpm): _____ Time Start: _____ Time End: _____

Final Parameters	Stabilization Guidance	Met?	Comments
pH		0.1	Y/N
Conductivity		3%	Y/N
Temp@		10%	Y/N
Final H2O level		feet	

O/G visible:	Y/N	Turbid?	Y/N
--------------	-----	---------	-----

Equipment Decontaminated: Y/N

Decontamination procedure used: _____

Signature: _____

Weather: _____

Newmont Mining Co
Cripple Creek & Victor Gold Mining Co

Surface Water Sampling Log

Location: Date:
Technician: Quarter:

Time	pH (S.U.)	Cond. (uS/cm)	Temp. (°C)	Notes
-	-	-	-	-

Sample
Method:

Oil/Gas visible [Y/N]

Turbid [Y/N]

Clear [Y/N]

Weather:

Signature:

Comments:

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Low-Flow Test Report:

Test Date / Time: 3/20/2024 2:36:24 PM
Project:
Operator Name:

Location Name: Device Location Initial Depth to Water: 93.15 ft	Estimated Total Volume Pumped: 12.784 gal Flow Cell Volume: 130 ml Final Flow Rate: 0.34 gal/min Final Draw Down: 2.22 ft	Instrument Used: Aqua TROLL 600 Serial Number: 1109809
--	---	---

Test Notes:
Grab sample, suspected broken casing

Low-Flow Readings:

Date Time	Elapsed Time	pH	Temperature	Specific Conductivity	RDO Concentration	Turbidity	ORP	Depth to Water	Flow
		+/- 0.1	+/- 3 %	+/- 3 %	+/- 10 %	+/- 10 %	+/- 10 %	+/- 0.33	
3/20/2024 2:36 PM	00:00	6.75 pH	7.85 °C	403.47 µS/cm	1.69 mg/L	43.88 NTU	9.1 mV	94.90 ft	0.34 gal/min
3/20/2024 2:37 PM	01:22	6.75 pH	6.56 °C	416.97 µS/cm	0.40 mg/L	43.68 NTU	50.5 mV	94.92 ft	0.34 gal/min
3/20/2024 2:42 PM	06:22	6.61 pH	6.08 °C	418.16 µS/cm	0.25 mg/L	50.58 NTU	104.2 mV	95.04 ft	0.34 gal/min
3/20/2024 2:47 PM	11:22	6.54 pH	6.25 °C	413.27 µS/cm	0.22 mg/L	59.01 NTU	95.7 mV	95.15 ft	0.34 gal/min
3/20/2024 2:52 PM	16:22	6.52 pH	6.40 °C	407.51 µS/cm	0.18 mg/L	61.13 NTU	77.7 mV	95.18 ft	0.34 gal/min
3/20/2024 2:54 PM	17:36	6.51 pH	6.37 °C	409.28 µS/cm	0.19 mg/L	59.00 NTU	83.9 mV	95.18 ft	0.34 gal/min
3/20/2024 2:59 PM	22:36	6.51 pH	6.30 °C	406.32 µS/cm	0.16 mg/L	59.11 NTU	112.1 mV	95.28 ft	0.34 gal/min
3/20/2024 3:04 PM	27:36	6.51 pH	6.39 °C	401.73 µS/cm	0.16 mg/L	60.35 NTU	83.7 mV	95.35 ft	0.34 gal/min
3/20/2024 3:09 PM	32:36	6.52 pH	6.42 °C	395.79 µS/cm	0.14 mg/L	60.92 NTU	57.2 mV	95.37 ft	0.34 gal/min
3/20/2024 3:14 PM	37:36	6.59 pH	7.44 °C	389.70 µS/cm	0.15 mg/L	67.32 NTU	79.1 mV	95.37 ft	0.34 gal/min

Samples

Sample ID:	Description:
GVMW15A	

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Appendix D
Reagent Water Specifications

REAGENT WATER SPECIFICATIONS

Quality Parameter	Type I	Type II	Type III
Bacteria, CFU/MI	10	1000	NA
pH	NA	NA	8-May
Resistivity, megohm-cm at 25° C	> 10	> 1	0.11
Resistivity, megohm-cm at 25° C	> 10	> 1	0.1

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Conductivity, umho/cm at 25° C	< 0.1	1	10
SiO ₂ , mg/L	< 0.05	< 0.1	< 1
Total Solids, mg/L	0.1	1	5
Total oxidizable organic carbon, mg/l	< 0.05	< 0.2	< 1

* NA = not applicable
Reference - Standard Methods for the
Examination of Water and Wastewater,
17th Edition, 1989.

APPENDIX E
WELL EVACUATION CALCULATION

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WELL EVACUATION CALCULATION - EXAMPLE

1. Measure depth to water from top of casing (TOC) prior to purging.
2. Begin purging well.
3. Determine well casing volume.

Total depth of well (TD) - depth to water (DTW) = total height of water in casing (H).

Example: TD = 66.60'

DTW = 46.15'

H = 20.45'

Volume in cubic feet (Vc) = $3.14 \times (\text{radius of well})^2 \times H$

Example: with a 4" casing, radius = 2" = 0.167'

$Vc = 3.14 \times (0.167')^2 \times 20.45'$

Vc = 0.09 square feet x 20.45 feet

Vc = 1.8 cubic feet

4. Convert cubic feet to gallons.

Cubic feet x 7.48 = gallons

Example: Vg = 1.8 cubic feet x 7.48 = 13.5 gallons

5. Three well volumes must be evacuated:

Example: $Vw_3 = 3 \times 13.5 \text{ gallons} = 40.4 \text{ gallons}$

6. Hence, to evacuate three well volumes in the above example, 40.4 gallons need to be purged before sampling.

Since all values in the above calculations are constant except for the height of water in casing (H), the constant values may be pre-calculated to simplify well volume determinations.

Thus:

For a 2" well, three well volumes (Vw_3) = H x 0.5

For a 4" well, three well volumes (Vw_3) = H x 2

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APPENDIX F
Low-Flow Purge Volume Calculation

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Groundwater Well - Field Volume Guide

Volume Equation:	$V(gal) = 0.1632 \cdot (r(in))^2 \cdot \Delta(ft)$
Delta (Δ) (ft):	$\Delta(ft) = \text{Stabilized Depth to Water}(ft) - \text{Initial Depth to Water}(ft)$

Required Pumping Volume: $Pumping\ Volume\ (gal) = \Delta V(gal) + Tubing\ Volume\ (gal)$

Diameter (in)	Volume (gal)
1"	$V(gal) = 0.0408 \cdot \Delta(ft)$
2"	$V(gal) = 0.1632 \cdot \Delta(ft)$
3"	$V(gal) = 0.3672 \cdot \Delta(ft)$
4"	$V(gal) = 0.6528 \cdot \Delta(ft)$
5"	$V(gal) = 1.02 \cdot \Delta(ft)$

ΔV - Volume b/w initial level and stabilized level (gal)					
Delta (Δ) (ft)	1" Well	2" Well	3" Well	4" Well	5" Well
0.1	0.0	0.0	0.0	0.1	0.1
0.3	0.0	0.0	0.1	0.2	0.3
0.5	0.0	0.1	0.2	0.3	0.5
0.7	0.0	0.1	0.3	0.5	0.7
0.9	0.0	0.1	0.3	0.6	0.9
1.1	0.0	0.2	0.4	0.7	1.1
1.3	0.1	0.2	0.5	0.8	1.3
1.5	0.1	0.2	0.6	1.0	1.5
1.7	0.1	0.3	0.6	1.1	1.7
1.9	0.1	0.3	0.7	1.2	1.9
2.1	0.1	0.3	0.8	1.4	2.1
2.3	0.1	0.4	0.8	1.5	2.3
2.5	0.1	0.4	0.9	1.6	2.6
2.7	0.1	0.4	1.0	1.8	2.8
2.9	0.1	0.5	1.1	1.9	3.0
3.1	0.1	0.5	1.1	2.0	3.2
3.3	0.1	0.5	1.2	2.2	3.4
3.5	0.1	0.6	1.3	2.3	3.6
3.7	0.2	0.6	1.4	2.4	3.8
3.9	0.2	0.6	1.4	2.5	4.0
4.1	0.2	0.7	1.5	2.7	4.2
4.3	0.2	0.7	1.6	2.8	4.4
4.5	0.2	0.7	1.7	2.9	4.6
4.7	0.2	0.8	1.7	3.1	4.8
4.9	0.2	0.8	1.8	3.2	5.0
5.1	0.2	0.8	1.9	3.3	5.2
5.3	0.2	0.9	1.9	3.5	5.4
5.5	0.2	0.9	2.0	3.6	5.6
5.7	0.2	0.9	2.1	3.7	5.8
5.9	0.2	1.0	2.2	3.9	6.0
6.1	0.2	1.0	2.2	4.0	6.2
6.3	0.3	1.0	2.3	4.1	6.4
6.5	0.3	1.1	2.4	4.2	6.6
6.7	0.3	1.1	2.5	4.4	6.8
6.9	0.3	1.1	2.5	4.5	7.0
7.1	0.3	1.2	2.6	4.6	7.2
7.3	0.3	1.2	2.7	4.8	7.4
7.5	0.3	1.2	2.8	4.9	7.7
7.7	0.3	1.3	2.8	5.0	7.9
7.9	0.3	1.3	2.9	5.2	8.1
8.1	0.3	1.3	3.0	5.3	8.3
8.3	0.3	1.4	3.0	5.4	8.5
8.5	0.3	1.4	3.1	5.5	8.7
8.7	0.4	1.4	3.2	5.7	8.9
8.9	0.4	1.5	3.3	5.8	9.1
9.1	0.4	1.5	3.3	5.9	9.3
9.3	0.4	1.5	3.4	6.1	9.5
9.5	0.4	1.6	3.5	6.2	9.7
9.7	0.4	1.6	3.6	6.3	9.9
9.9	0.4	1.6	3.6	6.5	10.1
10.1	0.4	1.6	3.7	6.6	10.3

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APPENDIX G
YSI Pro Water Quality Meter Calibration

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PURPOSE

CC&V uses the YSI Pro Plus water quality meter to collect water quality data that is submitted to regulatory agencies. It is the responsibility of the user to properly calibrate the probe before each use and keep an accurate record of each calibration.

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SCOPE

The YSI pro plus is to be calibrated prior to use for the day and details of the calibration are to be recorded on their calibration log sheetbook/ Sheet.

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YSI CALIBRATION

Dissolved Oxygen Calibration

Moisten the plastic cup by adding a small amount of clean water (1/8 inch) in the plastic storage cup or by moistening the sponge in the bottom of the cup. Make sure there are no water droplets on the DO membrane or temperature sensor. Then install the cup over the sensors screw it on the cable and then disengage one or two threads to ensure atmospheric venting. Make sure the DO and temperature sensors are not immersed in water. Turn the instrument on and wait approximately 5 to 15 minutes for the storage container to become completely saturated and to allow the sensors to stabilize.

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It is not necessary to calibrate in both % and mg/L or ppm. Calibrating in % will simultaneously calibrate mg/L and ppm and vice versa.

Press 'Cal' Highlight Probe ID

Highlight DO % and press enter to confirm.

Wait for the temperature and DO% values under "Actual Readings" to stabilize.

Then highlight Accept Calibration and press enter to calibrate. Or, press Esc to cancel the calibration.

Specific Conductance Calibration

Press 'Cal' Highlight Probe ID.

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After selecting the Probe ID, highlight Conductivity and press enter. Highlight the desired calibration method; Sp. Conductance. Remove sponge from cup. Place the sensor into a fresh, traceable conductivity calibration solution. The solution must cover the holes of the conductivity sensor that are closest to the cable. Ensure the entire conductivity sensor is submerged in the solution or the instrument will read approximately of half the expected value!

Choose the units in either SPC-us/cm and press enter.

Highlight Calibration value and press enter to input the value of the calibration standard. Then, once the temperature and conductivity readings stabilize, highlight Accept Calibration and press enter. Or, press Esc to cancel the calibration

pH Calibration

Press 'Cal'. Highlight Probe ID. After selecting your or Probe ID, highlight ISE (pH) and press enter. The message line will show the instrument is "Ready for point 1". The pH calibration allows up to six calibration points. Calibration for CCV purposes will be a 3 point calibration (4, 7, and 10). Place the sensor in a pH 7 buffer solution. The instrument should automatically recognize the buffer value and display it at the top of the calibration screen. If the calibration value is incorrect, the auto buffer recognition setting in the Sensor Setup menu may be incorrect. If necessary, highlight the Calibration Value and press enter to input the correct buffer value. Once the pH and temperature readings stabilize, highlight Accept Calibration and press enter to accept the first calibration point. The message line will then display "Ready for point 2".

To continue with the 2nd point, place the sensor in the second buffer solution (pH4). The instrument should automatically recognize the second buffer value (pH4) and display it at the top of the screen. If necessary, highlight the Calibration Value and press enter to input the correct buffer value. Once the pH and temperature readings stabilize, highlight Accept Calibration and press enter to confirm the second calibration point.

The message line will then display 'Ready for point 3' and you can continue with the 3rd calibration point (pH10) as detailed in previous steps for first and second calibration points.

Press 'Cal' to complete the calibration.

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Calibration Finalization

Upon completing the calibrations prior to field use, return probes to pH 4 storage solution. pH 4 is the recommended storage solution to prevent the dehydration of the pH probe. While using YSI, place pH 4 storage solution in the storage solution container that is maintained in case, to be used for continued storage at days end. Insure you have recorded the required values on the YSI Pro Plus calibration log.

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APPENDIX G

CC&V Sample Location Identified Sample Names

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Sample Location	Area	Sample ID Name
GV-06	Grassy Valley	GV-06
GVMW-25	Grassy Valley	GVMW-25
Seep-1	ECOSA Toe South	Seep-1
Seep-2	ECOSA Toe	Seep-2

APPENDIX H
Field Level Risk Assessment

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NEWMONT.
NORTH AMERICA

ZERO HARM!

Field Level Risk Assessment

Date: _____

Name: _____

Crew/Shift: _____

Foreman: _____

No Complacency: When I've done a task many times before, I will periodically review the steps and complete a risk assessment.

No Assumptions: I will not proceed with work until I know and understand the intent of everyone involved and ensure they know my intent.

Stay Focused: When I recognize my mind is not on the task, I will stop, take a breath, and fix my hazards before returning to work.

No Shortcuts: I will not allow myself to take the easy way by shortcutting procedures.

NEWMONT.
NORTH AMERICA

A Safety Journey

VITAL BEHAVIORS

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APPENDIX I
CC&V Surface Water Flume & Weir Calibration&
Groundwater SOP

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PURPOSE

Cripple Creek & Victor utilizes flumes and weirs to collect samples in a consistent and unbiased manner to characterize the actual quality of the water. This guide summarizes the current CC&V procedures that are established for water quality sampling, shipment, as well as the equipment used to collect samples and field parameters. The results from these samples are used to insure compliance with permits and regulatory agencies.

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SCOPE

This SOP outlines and covers the following

- Monitor pro and field log sheets
- Decontamination of sampling equipment
- Surface water sampling
- Ground water sampling, using both dedicated and submersible pumps
- Stream flow measurements and equipment
- Chain of Custody and sample shipment protocols

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Procedure

This procedure covers the sampling techniques that will be used to collect the monthly, bi-monthly, quarterly, and occasional spot sampling that will be required. Both surface water and ground water are outlined. Information that will assist in the sampling process is also located below, such as preparation, equipment information, decontamination, tablet based programs to assist in logging information, chain of custody and sample shipment protocol. The ground water sampling procedure outlined here is based on guideline outlined in the EPA low flow method.

For all field sampling procedures to occur both on and off mine site, property sampling technician should maintain safety as a priority when visiting sampling locations. At all times the technician should use their best judgement to determine if the sampling location is safe to perform the sampling task or inspection they are to perform. When visiting sampling locations there may be times when bad weather creates an unsafe condition for collecting samples or performing field inspections. When bad weather has caused a field location to be unsafe it is the technician's responsibility to not

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perform the task for that location. If a site is considered unsafe to collect a sample or perform an inspection the unsafe condition should be recorded on a field level risk assessment card, and if possible a picture taken with field equipment to be recorded within Monitor Pro 5 software.

Monitor pro

Tablets enabled with Monitor pro allow field samplers to collect sampling and field data directly to the web based database. One limitation with the monitor pro tablet configuration is that only the final parameter reading can be logged. It is suggested that a field log sheet is used to monitor the stabilization of parameters while sampling monitoring wells, to insure that stabilization has occurred.

Preparation

Prior to leaving office for field collect the needed pre-preserved bottles, sampling equipment, and decontamination equipment that will be needed for the planned sampling locations. By identifying and planning the sampling locations for the day you can minimize your vehicle load and maximize your potential sampling. Not all locations will require all equipment.

Decontamination Preparation

Three-phase decontamination is to be used. If decontamination solutions begin to appear dirty, the need to be changed out.

1. Prepare designated water containers with necessary water.
2. Prepare rinse containers for decontamination use, note which container is for Liquinox, H₂O, and D.I. H₂O
3. When ready to decontaminate fill rinse containers with appropriate H₂O
4. Add a small amount of nonphosphate detergent to the rinse container labeled "Liquinox"
5. Decontaminate submersible pump by submerging into rinse H₂O first, then submerge into "Liquinox" rinse container then submerge pump into D.I. H₂O rinse container. Submersible pump should remain in each container for approximately 1 minute. To decontaminate peristaltic pump purge approximately 250 ml of H₂O first through the line, then approximately 250 ml of "liquinox" through the line, then approximately 250 ml of D.I. H₂O through the line. To Decontaminate sounder submerge

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into rinse H₂O, then submerge in “Liquinox”, then submerge in D.I. H₂O. Sounder should remain in “liquinox” container for approximately 1 minute.

6. Properly dispose of H₂O used to decontaminate equipment

Surface Water Sampling

Prior to leaving for field identify the required parameters and analyses for planned sampling locations and collect equipment accordingly.

Surface Water Sampling Equipment List

- Field level risk assessment form (see section 4)
- YSI Professional Multi Meter, or Myron Ultrameter II
- Geo Peristaltic Pump
- D.I. water
- 100 ft. Tape measure surface water flow, this instruction is to aid in the inspections, and confirmation of
- Marsh McBirney 2000 portable Flow meter & Wadding Staff
- Wader Boots
- Monitor Pro compatible tablet
- Cooler with ice
- Sample Bottles
- 0.45 Micron Filters
- Sampling vessel
- Waterproof pen
- Pre preserved sample bottles
- Field sample log sheets

Stream Flow Measurements

Site inspection guidelines are as follows:

1. The channel should have as much straight run as possible.
2. Where the length of straight run is limited, the length up stream from the profile should be twice the downstream length.
3. The channel should be free of flow disturbances.
4. The flow should be free of swirls, eddies, vortices, backward flow, or dead zones.

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5. ~~Avoid areas immediately downstream from sharp bends or obstructions.~~
6. ~~If flume or weir is in place, use flow measurement from the device, insure the devices have been calibrated prior to leaving for field.~~

Marsh McBirney Flow Meter

~~Technical guidance can be found in the Instruction Manual for the model 2000 portable flow meter.~~

~~The method that is being used is the .4 method, or the 60% method.~~

1. ~~Upon arrival at field location perform field level risk assessment to determine location is safe to perform surface water sampling. Document risk assessment with field level risk assessment form, and record image with field equipment to be uploaded to Monitor Pro 5.~~
2. ~~Measure the width of the channel.~~
3. ~~Divide the width of the channel into a number of equal segments (more segments increases the accuracy of the flumes and weirs resultant flow). If the difference in mean velocity between two adjacent segments is greater than 10%, the segments should be smaller.~~

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SCOPE

~~Calibrations~~~~Work from the downstream side of the flumes and weir are to be conducted~~~~tape beginning on an annual basis and recorded in the proper shared~~~~drive location.~~

PPE

Safety Toe boots
Safety glasses
Hard hat
High-visibility vest

EQUIPMENT

Bucket
Stopwatch
Staff
Marsh-McBirney Flo-mate

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Level

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Flume Calibration

To confirm flume is level, use a long enough level to span the support bracketsbank side of the flume length wise, and across the width at the inlet, and the outlet. Check level regardless of absent flow.

If flume is out of level use the adjustment all threads and adjust flume to level. There are four adjustment rods for adjusting level. Some excavating may be necessary to access the adjustments.

If flow through flume is greater than 0.3 you may use flo-mate to calibrate flume. If flow is less than 0.3 use bucket and stop watch method (See Weir Calibration)

To calibrate flow through the flume, attach flo-mate reader to staff. Notestream, putting the Top-Setting Wading rod into the stream. Set the depth of water on flume staff.

4. Place staff and flo-mate in flume with instrument at 30% total the wading staff's sliding rod lock to the depth from of the water which is read at the bottom. Instrument will provide a of the rod.
5. Record the velocity and depth measurements.
6. Proceed to the next segments and make the depth measurement, adjustment.

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Calculate flow:

Depth x width x velocity = flow

7. Compare instrument Record the segment units used .5 ft., 1 ft., etc. then record the depth for that segment.

Flow is calculated with the continuity equation ($Q = \hat{a} \times A$)

Whereas Q is flow, \hat{a} is mean velocity and A is cross-sectional area. The flow of the channel is the sum of all segment areas times the mean velocity across the channel. Remember to account for the bank measurements.

Surface Water Sampling Procedure

1. Prior to leaving office turn on and calibrate YSI Pro plus water quality

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- meter, record in calibration log the pH buffer values, DO calibration value, and Specific Conductance Calibration Values (See YSI Calibration SOP).
2. ~~Upon arrival at field location perform field level risk assessment to determine location is safe to perform surface water sampling. Document risk assessment with field level risk assessment form, and record image with field equipment to be uploaded to Monitor Pro 5.~~
 3. ~~When at sampling location, place YSI probe directly in stream, allow parameters to stabilize and record parameters.~~
 4. ~~Collect data to calculate flow rate~~
 - a. ~~If location has a flume or weir, collect flow data from flume or weir and record~~
 - b. ~~If no flume or weir is present and sufficient flow is present collect flow data by the following stream flow measurement with known value attained from water depth in flume. Values procedure and the Marsh McBirney flow meter procedure. If insufficient flow is present to use Marsh McBirney flow meter estimate flow.~~
 5. ~~Collect a grab sample upstream from YSI probe by submerging a sampling vessel to mid depth of the stream, rinse the bottle 3 times with stream water and then collect sample.~~
 6. ~~Transfer sample into sample bottles that are labeled with the correct sample location, date, time, preservative and if filtration is required.~~
 - a. ~~To collect samples that do not require filtration, pour sample directly from sampling vessel.~~
 - b. ~~To collect samples that require filtration, use decontaminated peristaltic pump.~~
 - i. ~~Attach 45 micron filter to peristaltic pump tubing.~~
 - ii. ~~Allow sample to saturate and to flow through filter (approximately 200 mL).~~
 - iii. ~~Discharge and discard approximately 100 mL of H₂O from filter~~
 - iv. ~~Collect filtered sample in appropriate sample bottle(s).~~
 7. ~~Place samples in cooler on ice for storage.~~
 8. ~~Decontaminate sampling equipment.~~
 9. ~~Clean up sampling site and move to next location.~~

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Ground Water Sampling

~~Prior to leaving for field identify the required parameters and analyses for planned sampling locations and collect equipment accordingly.~~

Monitoring Well Sampling Equipment List

- ~~• Field level risk assessment form (see section 4)~~
- ~~• YSI Professional Multi Meter, or Myron Ultrameter II~~
- ~~• Peristaltic Pump~~
- ~~• D.I. water~~
- ~~• Monitor pro compatible tablet~~
- ~~• Cooler with Ice~~
- ~~• Sample Bottles~~
- ~~• 0.45 Micron Filters~~
- ~~• Sample Bucket~~
- ~~• Water level meters (150ft and 500ft)~~
- ~~• Geo Sub pump and controller~~
- ~~• Generator~~
- ~~• Appropriate keys for locks on the well head~~
- ~~• Dedicated pump controller~~
- ~~• Waterproof pen~~
- ~~• Field sample log sheets~~

Monitoring Well Sampling Procedure, Submersible Pump

~~Prior to leaving for field identify the required parameters and analyses for planned sampling locations and collect equipment accordingly.~~

- ~~1. Upon arrival at the well location, perform field level risk assessment to determine location is safe to perform surface water sampling. Document risk assessment with field level risk assessment form, and record image with field equipment to be uploaded to Monitor Pro 5.~~
- ~~2. With well location determined to be safe, complete a visual inspection of well vault / monument, casing, site condition and if a dedicated pump is present~~
- ~~3. Measure the static water level in the well and record measurement listed as (depth to water). Water levels should be measured with in 10% a precision of each other. Document values on ± 0.01 foot.~~
- ~~4. Decontaminate water level meter after use.~~

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5. You may calculate the flume calibration log. If volume of water in the well to aid in the purging process, there is no required purge volume for the EPA low flow is sampling standard, rather it is based on parameter stabilization.

a. 2 in = 617 ml/ft

b. 4 in = 2470 ml/ft

6. Insert decontaminated pump and tubing into the well descending slowly until pump is in the desired screened interval, avoid letting the pump come in contact with the well bottom.

7. Attach the discharge tubing to the bottom port on the YSI water quality meter flow cell.

8. Connect the pump to the controller with the cord provided

9. Start generator, allow it to warm up.

10. Connect controller box to generator with power cord provided.

11. Turn pump controller on and begin to purge water.

a. When flow is present adjust flow rate by using the up and down arrow buttons.

b. Attempt to maintain a flow rate of to 0.1 to 0.5 L of water per minute and generally speaking not sufficient to use flo-mate, use calibrated bucket to exceed 1 L per minute.

c. Use 1 L vessel and stop watch to measure flow rate and record.

12. Begin to record parameters and draw down to monitor stabilization every 3 to 5 minutes, including;

a. pH

b. Specific conductance

~~Dissolved Oxygen~~

Weir Calibration

c. Use calibrated bucket

d. Temperature

13. Continue to monitor purge flow rate.

14. While the well is purging, compile and label sample bottles.

15. Three successive parameter readings must be within the following tolerances of each other in order to collect samples.

a. pH +/- 0.1

b. Specific Conductance +/- 3%

c. Dissolved Oxygen +/- 10%

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- d. Temperature +/- 10%
- 16. Upon stabilization of parameters collect samples in pre-labeled and pre-preserved bottles, when possible following this general order; samples should be collected without passing through YSI flow cell.
 - a. Total Metals (unfiltered)
 - b. Dissolved Metals (filtered)
 - c. Cyanide
 - d. TSS
 - e. Nitrate/Nitrites
- 17. To collect filtered samples;
 - a. Attach 45 micron filter to peristaltic pump tubing.
 - b. Allow sample to saturate and to flow through filter (approximately 200 mL).
 - c. Discharge and discard approximately 100 mL of H₂O from filter
 - d. Collect filtered sample in appropriate sample bottle(s).
- 18. Place samples in cooler on ice.
- 19. Turn off pump at pump controller box, remove power cord from pump.
- 20. Pull pump and tubing from well, reeling pump cord back onto pump reel.
- 21. Decontaminate equipment (submersible pump, peristaltic pump)
- 22. Clean up sampling site and move to next location.

Monitoring Well Sampling, Dedicated Electric Pumps

The compliance well at the External Storage Pond (ESPMW) has a 240 volt dedicated pump. Grounding on all electrical driven pumps must be certified by CC&V's electrical department once per year. It is of utmost importance that the sampler inspects all electrical leads and connection prior to use. The Onan 4000 watt generator has adequate power to run this pump. After the generator has warmed up, plug the twist lock power lead into the generators 30 amp receptacle, and switch the voltage control to 240 volts. There is a short time delay before water use is pumped up, and out the discharge hose. If the well is purged dry the flow controller will shut the pump off automatically. There are two wells in Grassy Valley, four Wells in Squaw Gulch, and four wells in lower Arequa Gulch that are also dedicated electric pumps that require a generator, and either the BCD, or converter controller. Be familiar with the

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~~safety features with the generator, and have the electrical department approve all grounding requirements prior to use.~~

- ~~1. Upon arrival at well location perform field level risk assessment to determine location is safe to perform surface water sampling. Document risk assessment with field level risk assessment form, and record image with field equipment to be uploaded to Monitor Pro 5.~~
- ~~2. With well location determined to be safe, complete a visual inspection of well vault / monument, casing, site condition and if a dedicated pump is present. Upon arrive at the well site, complete a visual inspection of well vault / monument, casing, site condition and if a dedicated pump is present.~~
- ~~3. Measure the static water level in the well and record measurement listed as (depth to water). Water levels should be measured with a precision of +/- 0.01 foot.~~
- ~~4. Decontaminate water level meter after use.~~
- ~~5. You may calculate the volume of water in the well to aid in the purging process, there is no required purge volume for the EPA low flow sampling standard, it is based on parameter stabilization.~~
 - ~~a. 2 in = 617 ml/ft~~
 - ~~b. 4 in = 2470 ml/ft~~
- ~~6. Connect tubing to sampling port on monument/ well vault.~~
- ~~7. Attach the discharge tubing to the bottom port on the YSI water quality meter flow cell.~~
- ~~8. Connect the pump to the controller with the cord provided.~~
- ~~9. Start generator, allow it to warm up.~~
- ~~10. Connect controller box to generator with power cord provided.~~
- ~~11. Turn pump controller on and begin to purge water.~~
 - ~~a. When flow is present adjust flow rate by using the up and down arrow buttons.~~
 - ~~b. Attempt to maintain a flow rate of to 0.1 to 0.5 L of water per minute and generally speaking not to exceed 1 L per minute.~~

~~c. Use 1 L vessel and stop watch to time the collection of water to calculate flow measure flow rate and record.~~

~~Volume of bucket / Collection time= flow x 60= GPM~~

~~Ex. 1 gallon 5.0 seconds~~

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1/5 x60 =12 GPM

- ~~12. Compare calculated flow to flow reading from weir. Begin to record parameters and draw down to monitor stabilization every 3 to 5 minutes, including;~~
 - ~~a. pH~~
 - ~~b. Specific conductance~~
 - ~~c. Dissolved Oxygen~~
 - ~~d. Temperature~~
- ~~13. Continue to monitor purge flow rate.~~
- ~~14. While the well is purging, compile and label sample bottles.~~
- ~~15. Three successive parameter readings must be within the following tolerances of each other in order to collect samples.~~
 - ~~a. pH +/- 0.1~~
 - ~~b. Specific Conductance +/- 3%~~
 - ~~c. Dissolved Oxygen +/- 10%~~
 - ~~d. Temperature +/- 10%~~
- ~~16. Upon stabilization of parameters collect samples in pre-labeled and pre-preserved bottles, when possible following this general order; samples should be collected without passing through YSI flow cell.~~
-
- ~~17. To collect filtered samples;~~
 - ~~a. Attach filter to sample tube.~~
 - ~~b. Allow sample to saturate and to flow through filter (approximately 200 mL).~~
 - ~~c. Collect filtered sample in sample bottle labeled for filtration.~~
- ~~18. Place samples in cooler on ice.~~
- ~~19. Turn off pump at pump controller box, remove power cord from pump, unplug controller from generator.~~
- ~~20. Clean up site and move to next location~~

Chain-of-Custody and Sample Shipment

~~A chain-of-custody (COC) is a procedure designed to allow the operator to reconstruct how and to whom the sample is transferred.~~

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1. ~~Upon preparing samples for shipment, the sampler should complete a COC.~~
2. ~~Insure all bottles are labeled with the correct time, date, analyte, preservative, required filtration, and location of the sample by comparing to field notes.~~
3. ~~List sample location, date, time, field parameters, number of containers, types of preservatives present and the requested analytes on the chain of custody form, print form (2 copies).~~
4. ~~Double check the sample bottle labels against the chain of custody.~~
5. ~~Print the pre filled ground water or surface water analytical request form, dependent on types of samples being submitted.~~
6. ~~Place COC and analytical request form in plastic bag inside the cooler to be shipped.~~
7. ~~After inspection of samples and comparison to COC, place samples in cooler, along with adequate contained ice to keep samples with in the hold temperatures during shipment.~~
8. ~~If needed use packing material to fill spaces in cooler that contained ice too large to fill.~~
9. ~~Close cooler lid and use 2 inch clear packaging tape to secure both ends of cooler, wrap tape multiple times around the ends of cooler, followed by strapping tape.~~
10. ~~Attach a preprinted shipping label to the top of cooler, place shipment identification sticker on copy of COC.~~
11. ~~File copy of COC until lab returns results and original COC.~~

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Risk Assessment	
What task am I doing?	
What could change/go wrong? (People or Planet)	
What can I do about it? Is there a better way? SDS?	
Foreman Comments?	

Risk Assessment	
What task am I doing?	
What could change/go wrong? (People or Planet)	
What can I do about it? Is there a better way? 505?	
Foreman Comments?	

NEWMONT.
NORTH AMERICA

ZERO HARM!

Field Level Risk Assessment

Date: _____

Name: _____

Crew/Shift: _____

Foreman: _____

No Complacency: When I've done a task many times before, I will periodically review the steps and complete a risk assessment.

No Assumptions: I will not proceed with work until I know and understand the intent of everyone involved and ensure they know my intent.

Stay Focused: When I recognize my mind is not on the task, I will stop, take a breath, and fix my hazards before returning to work.

No Shortcuts: I will not allow myself to take the easy way by shortening procedures.

NEWMONT.
NORTH AMERICA

a Safety Journey

VITAL BEHAVIORS

Notes/Sketch Pad

Work Area Inspection			
Work Area 1:			
Work Area 2:			
Work Area 3:			
Describe corrective actions on page 2:	Yes or No or N/A		
	W/A 1	W/A 2	W/A 3
Do I have the proper PPE?			
Vehicles secured from movement?			
Safe access to all areas?			
Equipment inspected?			
Emergency Muster/Evac. Points?			
Fire Extinguishers & Location?			
Energy Isolated - Elec/Mech/Hyd?			
Ventilation?			
Guards and Barriers in Place?			
Ground Conditions Inspected?			
Electrical Cords and Ground Rods?			
Area Housekeeping Good?			
Flammable Storage & Labeling?			
Proper Tools and Condition?			
Pre-Op Inspections Completed?			
Am I in the Line of Fire?			
Lifting Gear, Slings & Hooks Good?			
Hazards Corrected or Barricaded?			
Oil Spills or Leaks Properly Contained?			
Adequate Lighting for Job?			
Work Place Exam Completed?			
Any Work Permits Needed? (i.e. Hot Work, Confined Space, Excavation/Trenching, Critical Lift, etc.)			

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APPENDIX J
CC&V Sample Shipping SOP

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CC&V – Sample Prep & Shipping SOP

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SAMPLE PREP & SHIPPING SUMMARY

The following inventory of prepared sample bottles must be continually kept up:

- 1) **Cyanide Bottles:** green labels, brown 125 ml bottles preserved with 0.2 ml of Sodium Hydroxide ~60 bottles
- 2) **Metals Bottles:** red labels, white 250 ml bottles preserved with 1.5 ml of Nitric Acid ~ 60 bottles
- 3) **Nitrogen Bottles:** yellow labels, white 250 ml bottles preserved with 0.5 ml of Sulfuric Acid ~40 bottles
- 4) **Mineral Bottles:** ~~black~~white labels, 500 ml bottles non-preserved ~ 60 bottles

5) **Hexavalent Chromium Kit:** green labels, 250 mL bottles preserved with buffer, NaOH for pH adjustment, and pH strips ~ 20 bottles

6) **Sulfide Bottles:** purple labels, 250 mL bottles preserved with Zinc Acetate NaOH ~ 20 bottles

5) 5) **Verify expiration dates** on reagent used in bottle preservation

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Before running too low on bottles or preservatives order from SVL. Always confirm sample supplies before each new quarter to ensure sufficient supply for quarterly compliance sampling. -Analytical by emailing SVLSupplies@svl.net

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ALWAYS WEAR NITRILE GLOVES AND SAFETY GLASSES WHEN PRESERVING BOTTLES!!!!

Shipping Samples

Check with staff daily if there will be samples ready to ship out at the end of the day, if there are samples to ship follow the shipping instructions:

- 1) Match COC's to bottles in the refrigerator
- 2) Sign and date the COC's
- 3) Make copies of COC's, place original in plastic bag – these will go with the cooler with samples. Place copies in red file folder for reference.
- 4) Load up samples, COC's, and ice into “five-day” cooler.
- 5) Sign and date Custody seal and place vertically across lid and cooler.
- 6) Tape up cooler lengthwise and width wise
- 7) Take cooler ~~and mailing with appropriate shipping label up to and drop off at the Post Office~~ UPS Store in Woodland Park, CO no later than 4:15 p.m. ~~30 PM~~
- 8) Get mailing label sticker with the tracking number and place it on the copies of the COC's
- 9) Keep freezer stocked full of ice!!

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Procedures for Water Quality Sample Shipments.

- 1) Any water quality samples collected from Monday thru Thursday should be shipped either before or on Thursday.
 - a) CDPS Water Quality Samples are to be shipped the day the sample was collected if time allows, otherwise they must be shipped the next day.
 - b) Ground & Surface Water Samples are to be shipped either on the day the sample was collected or the next day.
 - c) VLF Water Quality Samples are to be shipped either on the day the sample was collected or within two days.
- 2) Any water quality sample collected on Friday may sit in the refrigerator until the following Monday, but must be shipped on that Monday.
 - a) No water quality samples are to sit in refrigerator over a weekend, without prior approval.

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APPENDIX K
Myron L II Ultrameter Calibration procedure

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~~3.1 Specific Conductance Calibration 63~~

~~3.2 pH Calibration 63~~

~~3.3 Total Dissolved Solids Calibration Error! Bookmark not defined.~~

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PURPOSE

The Myron L II is used to collect data that is submitted to regulatory agencies. It is the responsibility of the user to ensure that the probe is properly calibrated and that an accurate record is kept with each calibration.

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SCOPE

The Myron L II is to be calibrated prior to use for the day and details of the calibration are to be recorded in calibration log [sheetbook/ Sheet](#).

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Myron L II Calibration

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Specific Conductance Calibration

- Push the COND button
- With KCL 7000 solution, fill the conductivity cell cup completely and dump it 2 times.
- With KCL 7000 solution, fill the conductivity cell cup completely a 3rd time, and let it settle on a reading.
- Push the CAL button.
- Use the up or down buttons to make the display read 7000.
- Hit the CAL button, to accept the calibration.
- If the display does not read 7000, repeat steps 4-6 until it does.

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Total Dissolved Solids Calibration

- Push the TDS button
- With the 442-3000 solution, fill the conductivity cell cup completely and dump it 2 times.
- With the 442-3000 solution, fill the conductivity cell cup completely a 3rd time and let it settle on a reading.
- Push the CAL button.
- Use the up or down buttons to make the display read 3000.
- Hit the CAL button, to accept the calibration.
- If the display does not read 3000, repeat steps 4-6 until it does.

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PH Calibration

- Push the PH button
- Remove the protective rubber cap from the PH sensor well.
- With the PH7 solution, fill the PH sensor well and the conductivity cell cup completely and dump it 2 times.
- With the PH7 solution, fill the PH sensor well and the conductivity cell cup completely a 3rd time and let it settle on a reading.
- Push the CAL button.
- Use the up or down buttons to make the display read 7.0.
- Hit the CAL button, to accept the calibration.
- If the display does not read 7.0, repeat steps 5-7 until it does.
- Repeat steps 3-8 with the PH4.0 and the PH10.0 solutions.

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APPENDIX L
Aqua Troll 600 Calibration procedure

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